The paper discusses various ways of activating phosphorus slag by introducing additives for the development of phosphorus slag binders (PSB), replacing cement. Considering that pseudowollastonite is the main mineral of phosphorus slag and without activating components does not possess the binding properties necessary for the production of building materials based on them, we used compositions of small amounts of sodium hydroxide with alkali metal salts, the anions of which form poorly soluble compounds with calcium. When choosing activating components, scarce alkaline additives were replaced by waste from chemical plants, which allows a passing solution of their practical use and environmental problems.

The strength at a sodium hydroxide content of 1–4 % after curing of slag samples of various batches was in the range of 50.0–70.0 MPa. Samples of binders of normal hardening at the age of 28 days with a sodium hydroxide content of 0.5; 1.0, 2 and 4 % had the strength of 20.3; 35.4; 45.6; 55.8 MPa, respectively. The effect of the combined presence of alkali and salt is especially noticeable for small amounts of sodium hydroxide. Binders containing a composition of cement with salts under normal conditions and after curing showed a slightly lower strength than in an alkaline medium. With a constant cement content (4 %), the strength indicators increase with an increase in the proportion of the salt additive, reaching at 4 % its maximum value. The effect of the nature of activators on pH was determined. The data obtained indicate the advantages of using PSB and various industrial wastes with a low content of alkaline compounds in the production of building materials based on them.

Keywords: phosphorus slag, cement, sodium hydroxide, phosphorus slag binders, construction materials

UDC 69:624.01:691:624.01:691
DOI: 10.15587/1729-4061.2021.242814

DEVELOPMENT AND RESEARCH OF THE INFLUENCE OF THE COMPOSITION AND CONCENTRATION OF ACTIVATORS ON THE STRENGTH OF PHOSPHORUS SLAG BINDERS

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Received date 09.09.2021
Accepted date 19.10.2021
Published date 29.10.2021

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1. Introduction

Granular phosphorus slags are by-products (man-made) obtained in the production of phosphorus by the method of sublimation in electrothermal furnaces. Their granulation is carried out at a temperature of about 1,450 °C by draining the fiery liquid slags into a pool of water.

Phosphorus and its compounds (phosphorus fertilizers) are mainly obtained from phosphorite (calcium salt of phosphoric acid $\text{Ca}_3(\text{PO}_4)_2$ or $3\text{CaO}_2\text{P}_2\text{O}_5$), the richest deposits of which in Kazakhstan are located in the Karatau Mountains.
In electric furnaces at a temperature of about 1,450 °C, the following reaction takes place in the corresponding mixtures of phosphorite (3CaO\(\cdot\)P\(_2\)O\(_5\)), sand (SiO\(_2\)) and carbon (C):

\[
\text{Ca}_3(\text{PO}_4)_2+3\text{SiO}_2+5\text{C} \rightarrow 3\text{CaO}+3\text{SiO}_2+2\text{P}^\circ+5\text{CO}.
\] (1)

The chemical composition of phosphorus slags is characterized by constancy. The binding properties of monominerals and glasses in the composition of phosphorus slag were studied by [1, 2]. The products of the hydration of \(\alpha\)-CaO\(\cdot\)SiO\(_2\) with Ca(OH)\(_2\) are fibrous calcium hydroxylates of the type CSN (1) and high-base calcium hydroxylate-a-hydrate C\(_2\)S [1–3]. It was found that pseudowollastonite, which is the main mineral of phosphorus slags without activating components does not have astringent properties, but in the presence of calcium hydroxide it acquires them and is able to harden independently.

Three methods have been developed for the purification of granular slags from harmful impurities of phosphine and hydrogen fluoride: liquid-phase, thermal and mechanochemical. The most common is the mechanochemical method of cleaning. Mechanochemical purification of granular phosphorous slags is based on the reactions of phosphine oxidation and neutralization of hydrogen fluoride by introducing oxidizing and neutralizing additives into the slags during their drying and grinding [4].

The production technology of phosphorus slag binders is based on theoretical assumptions about the destruction of silicon- and calcium-oxygen polyhedral groups of slag under the influence of various activating components [2, 3].

The activating components of the slag binder, depending on their nature and the mechanism of action on the hardening system, are divided into six groups [5]:

1. Caustic alkalis (ROH).
2. Unsilicate salts of weak acids (R\(_2\)CO\(_3\), R\(_2\)SO\(_3\), R\(_2\)S, RF).
3. Silicate salts (R\(_2\)O [0.5–4.0] \(\cdot\)SiO\(_2\)).
4. Aluminate salts (R\(_2\)OAl\(_2\)O\(_3\)).
5. Aluminosilicate salts (R\(_2\)OAl\(_2\)O\(_3\)SiO\(_2\)).
6. Non-silicate salts of strong acids (RCI, R\(_2\)SO\(_4\)) [2–5].

Therefore, studies devoted to the activation process of an inert phosphorus slag using a composition of small amounts of sodium hydroxide with alkali metal salts, the anions of which form poorly soluble compounds with calcium, are of scientific significance.

### 2. Literature review and problem statement

In [6], studies are devoted to the experimental assessment of the effect of activators on the strength properties of slag concretes during their hardening under conditions of steaming and autoclaving. As activators, sulfates and chlorides of alkali and alkaline earth metals Na\(_2\)SO\(_4\), K\(_2\)SO\(_4\), MgSO\(_4\), mirabilite, as well as chlorides of alkali and alkaline earth metals MgCl\(_2\), CaCl\(_2\), carnalite, natural bischofite are used. It is shown that with the introduction of chlorides to activate the slag in an amount of 1 % of the slag weight and hardening of concrete under conditions of elevated temperatures and steam pressure, high-strength concretes with strength from 58 to 80 MPa are obtained. It is important to note that salts containing MgCl\(_2\), CaCl\(_2\) chlorides, carnalite, natural bischofite have a negative effect on corrosion resistance and require additional introduction of inhibitors, the results of which are not presented by the authors.

The authors [7] propose additives – APEG-type PCE and HPEG-type PCE, which can effectively improve the fluidity of the slag. The work is devoted to the selection of additives to reduce the increased viscosity, due to the activation of NaOH with a high concentration. From an economic point of view, the recommendations by the authors on the parallel introduction of two types of additives, in terms of functionality, that reduce the effectiveness of one of them are unprofitable.

The paper [8] publishes data on the technical characteristics of phosphorus slags: composition, activity, safety levels, utilization, as well as the results of testing hybrid cements, activated with alkali, containing phosphorus slag in the binder:

1. C. Portland cement clinker;
2. PS. phosphorus slag;
3. G. gypsum; CG. calcined gypsum;
4. NS. sodium sulfate;
5. CAS. calcined alumina.

It was noted that they showed improved stability compared to Portland cement samples when exposed to sulfate solutions. The addition of phosphorus slag also reduced the tendency for harmful alkali reaction with the unit. The work [8] shows the safe use of alkaline activators in metal aggregates used in the production of concrete mixtures based on phosphorus slag.

The work [9] experimentally studied the effect of 5–7 % alkali addition and sodium silicate modulus in the activator solution on workability, compressive strength and shrinkage during drying of slag concrete. In the test results, in most mixtures, due to an increase in the modulus of sodium silicate and the concentration of an alkaline solution: sediment, compressive strength and shrinkage rate during drying of the samples were increased, but it is necessary to note a reduction in the setting time, which leads to disruption of the technological process.

The objectives of the study [10] were to obtain a binder based on phosphorus slag and to study the suitability of polymeric ore tailings containing calcium carbonate and sodium barium compounds, and their combination as activators, by studying the effect of the type and dosage of activators on compressive strength. This work shows the promise of expanding the area of using waste with an alkaline reaction for the activation of phosphorus slag.

In [11], a statistical experimental design was applied to predict and optimize the compressive strength of alkali-activated phosphorus slag at different ages (3, 7, and 28 days). For this, samples of a binder were prepared with different molar ratios SiO\(_2\)/Na\(_2\)O (S/N), Na\(_2\)O/Al\(_2\)O\(_3\) (Na/Al) and H\(_2\)O/Al\(_2\)O\(_3\) (H/Al) as alkaline activators. The optimal chemical composition of the activator for obtaining maximum compressive strength was achieved as 0.39, 1.34 and 30 for molar ratios S/N, Na/Al and H/Al, respectively, with compressive strength of 30, 65 and 100 MPa through 3, 7 and 28 days of curing. As can be seen from the given strength data, the use of alkaline activators is highly effective even without heat treatment. The use of pure chemical compounds with a high content of alkaline activators is justified only for laboratory research and is excluded in a market economy.

The study [12] is devoted to the technical properties of alkali-activated slags (AAS) obtained by mixing an activator consisting of sodium silicate and sodium hydroxide at room temperature. AAS pastes were prepared using a mixture of activators with five different silicate modules 1,
1.2, 1.4, 1.6, and 1.8 and 3 different Na₂O contents: 6 %, 8 %, and 10 %. The authors obtained good workability, with a silicate modulus of 1 and a Na₂O content of 8 % at room temperature, high compressive strength of 55 MPa and higher, which is explained by the increased total alkali content. It is shown [12] that for all AAS pastes, a fast setting from 9 to 36 minutes, an increase in drying shrinkage with an increase in the silicate modulus are not applicable to the production process.

The paper [13] examines the effect of curing parameters such as pre-cure time, cure time and temperature on the compressive strength of a phosphorous slag cement mortars activated with two different compound activators, including NaOH + Na₂CO₃ and Na₂CO₃ + Ca(OH)₂, as well as the microstructure, morphology of solidified solutions. It is important to note that the use of high-temperature curing (autoclave t=175 °C) or a combined mode of low-temperature and high-temperature leading to a threefold increase in compressive strength compared to curing conditions in the environment is inexpedient in terms of energy costs and economically unfavorable for large-scale construction.

In [14], the effect of phosphorus slag from 10 % to 50 % (by weight) on the setting time and water demand of cement pastes of normal consistency, fluidity, resistance to carbonization and compressive strength of cement slurries was investigated. The compressive strength of the cement slurry was also reduced by the phosphorus slag from 10 % to 50 %, and based on the compressive strength of the cement slurry, it was concluded that the activity of the phosphorus slag was low, which is confirmed by its structural feature [1].

Considering that phosphorus slags are extremely valuable raw materials used as the main component or mineral additive in the production of clinker-free (slag-alkali) binders, slag Portland cement and Portland cement, the papers [15, 16] consider all known structural and physical and mechanical characteristics of granular phosphorus slags, including phase, as well as effective ways to neutralize them from harmful gases (PH₃, H₂S, HF, etc.) by reviewing the known theoretical and experimental studies.

As the results of the above studies show, in order to obtain active binders and improve the physical and mechanical properties of granular phosphorus slags, taking into account their inertness, alkaline additives (in the form of pure chemical compounds) are mainly proposed, obtained separately or more often by mixing alkaline activators from sodium silicate and sodium hydroxide with a total content of Na₂O in the mixture of 10 % and more, which provide high strength indicators and lead to increased viscosity, which affects the reduction of the setting time and excludes the introduction of a production technological process.

It is important to note that the use of high-temperature curing (autoclave t=175 °C) or a combined mode of low-temperature and high-temperature leading to a threefold increase in compressive strength compared to curing conditions in the environment is inexpedient in terms of energy costs and economically unavailable for large-scale construction. It is known that salts containing MgCl₂, CaCl₂ chlorides, carnalite, natural bischofite negatively affect corrosion resistance and require additional introduction of inhibitors.

3. The aim and objectives of the study

The aim of the study is the development of highly active phosphorus slag binders of hydrothermal and normal hardening at low consumption of deficient alkaline additives [16].

To achieve the aim, the following objectives were set:
- selection of the composition of small amounts of sodium hydroxide with pure alkali metal salts, the anions of which form poorly soluble compounds with calcium;
- selection of a composition of small amounts of sodium hydroxide with waste containing compounds corresponding in composition to model salts of alkali metals, the anions of which form poorly soluble compounds with calcium;
- study of the effect of sodium hydroxide, salts, depending on their nature, concentration on strength under normal conditions;
- study of the effect of cement, salts, depending on their nature, concentration on strength under normal conditions.

4. Materials and methods of research

As binders, phosphorus slag binders (non-annealed binders) obtained by grinding granulated electrothermophosphorus slag (Taraz, “United Phosphorus”, Republic of Kazakhstan), with activating components, neutralizing additives or without them, depending on the field of application, were used. In the second case, the activation of the slag occurs in the concrete mixture by activating components supplied to the concrete mixer in a dissolved state with mixing water (for example, salt) or in a solid state through a dispenser (cement) (Table 1).

<table>
<thead>
<tr>
<th>Material</th>
<th>Chemical composition, wt. %</th>
</tr>
</thead>
<tbody>
<tr>
<td>Electrothermophosphorus slag</td>
<td>SiO₂</td>
</tr>
<tr>
<td></td>
<td>42.68</td>
</tr>
</tbody>
</table>

Cement of the Kant cement plant (Kant, Kyrgyz Republic), sodium hydroxide, and salts were used as activators. The salts used were sulfate, sulfite, carbonate, phosphate, and sodium oxalate, potassium fluoride, and a synthetic soda-sulfate mixture (SSM) containing 30 % soda and 70 % sodium sulfate. The salts used by us are able to activate phosphorus slag in autoclave conditions, allowing us to obtain binders with the strength in the test in the range of 60–90 MPa [4]. But when steaming and under normal conditions, only salts that hydrolyze in an aqueous medium to form an alkali, such as phosphates and carbonates of alkali metals, have an activating ability.

The samples were formed with the dimensions of 4×4×16 cm from ground slag with a specific surface area of 300–350 m²/kg, closed with a solution of alkali and salt at V/V=0.21–0.22. Compaction of the mixtures was carried out on a standard vibration platform for 650 s, steaming – at 95 °C according to the regime of 2+7+2 h. Samples that hardened under normal conditions were stored under water in hermetically sealed baths. The strength test of the binders was carried out according to GOST 310.4-81 “Cements. Methods of bending and compression strength determination” [17] (Table 2).
Technology organic and inorganic substances

Table 2

<table>
<thead>
<tr>
<th>Indicator</th>
<th>Normative document</th>
<th>Actual value</th>
</tr>
</thead>
<tbody>
<tr>
<td>Fineness of grinding through the sieve No. 008, %</td>
<td>At least 85</td>
<td>93.16</td>
</tr>
<tr>
<td>Normal density, %</td>
<td>–</td>
<td>24.75</td>
</tr>
<tr>
<td>Setting time, hour-min:</td>
<td></td>
<td></td>
</tr>
<tr>
<td>start</td>
<td>not earlier than</td>
<td>0–50</td>
</tr>
<tr>
<td>end</td>
<td>no later than</td>
<td>7–53</td>
</tr>
<tr>
<td>Strength at the age of 28 days, kgf/cm²:</td>
<td></td>
<td></td>
</tr>
<tr>
<td>bending</td>
<td>At least 8,554</td>
<td>60.8</td>
</tr>
<tr>
<td>compression</td>
<td>At least 85,392</td>
<td>402.3</td>
</tr>
</tbody>
</table>

As a fine aggregate, feldspar sands of the Nikolaevsky and Kapchagay deposits of the Republic of Kazakhstan were used (Table 3).

Table 3

<table>
<thead>
<tr>
<th>Properties</th>
<th>Unit of measure</th>
<th>Nikolaevsky quarry (Almaty region, Republic of Kazakhstan)</th>
<th>Kapchagay quarry (Almaty region, Republic of Kazakhstan)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Bulk density</td>
<td>kg/m³</td>
<td>1.510</td>
<td>1.515</td>
</tr>
<tr>
<td>True density</td>
<td>kg/m³</td>
<td>2.67</td>
<td>2.60</td>
</tr>
<tr>
<td>Content of dust and clay particles</td>
<td>wt. %</td>
<td>2.0</td>
<td>2.9</td>
</tr>
</tbody>
</table>

The specific surface area of ground slag and slag binder was not less than 300 m²/kg or not more than 5 % of the residue on the sieve N 008.

For the study, 2 systems were taken:
1) “slag-sodium hydroxide-salt”; 2) “slag-portland cement-salt”.

In the course of the study, the influence of cement, sodium hydroxide, and salts, depending on their nature and concentration, on the strength was studied.

The activity (grade) of the slag binder was determined similarly to the activity of cement GOST 310.4–81 “Methods of determining, on the strength was studied. For example, with the addition of 1.0 % of activator, it was equal to 3.4; 5.5; 9.11 and 13.2 MPa [18].

Cinder-alkaline binders consist of neoplasms in the form of sodium-calciunm hydroxysilicates or hydroaluminosilicates [5]. A method for activating blast furnace-slag binders with small amounts of alkaline compounds was proposed by [17], but was not widely developed. It is known [4] that during the hydration of slag under autoclave conditions in the presence of small amounts of activators, calcium hydrosilicates of type CSH(B) or tobermorite are formed with the formation of an equally strong cementing bond. In order to accelerate the process of transformation of vitreous monobasic calcium silicate into hydrated compounds, studies were made on the effect of activators on strength were carried out under normal conditions (18–23 °C after 3, 7, 28 days) and under curing conditions at 95 °C according to the 2+7+2 h regime. The kinetics of alkaline activation of phosphorus slag cement according to the data of compressive strength was studied [6].

The temperature of the slag samples from different batches was in the range of 50.0–70.0 MPa with an alkali content of 1–4 %.

In the presence of 0.5 %, the strength was lower (25–30 MPa). Samples of binders of normal hardening at the age of 28 days with a sodium hydroxide content of 0.5; 1.0; 2 and 4 % had the strength of 20.3; 35.4; 45.6; 55.8 MPa, respectively, at the age of 7 days with the same amounts of activator, it was equal to 3.4; 5.5; 9.11 and 13.2 MPa [18].

When assessing the activity of binders by the strength of mortar samples on standard sand and freshly obtained slag (composition 1:3 at V/V=0.36), it was found that their strength in the presence of 1–2 % sodium hydroxide by weight of the slag after curing is 20.0–30.0 MPa. Low-alkali solutions on PSB are characterized by a slow strength gain in the early stages of hardening, especially for mature slags [19].

Thus, small amounts of alkalis (1–4 %) allow us to obtain PSB with sufficient activity. The effect of the combined presence of alkali and salt is particularly noticeable for small amounts of sodium hydroxide.

It was found that in an alkaline environment, sodium sulfate dramatically increases the activity of binders under curing conditions. For example, with the addition of 1.0 % alkali and 1–2 % salt, the strength of the samples was 67.0–69.0 MPa after curing. At 4 % sodium sulfate, the strength of the samples did not increase in comparison with binders with a lower salt content (Table 4).

Table 4

<table>
<thead>
<tr>
<th>Amount of activators, wt. %</th>
<th>Sodium hydroxide</th>
<th>Bending strength, MPa</th>
<th>Compressive strength, MPa</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Salt</td>
<td>Normal conditions</td>
<td>r=95 °C, 2+7+2 h</td>
</tr>
<tr>
<td></td>
<td></td>
<td>3 days</td>
<td>7 days</td>
</tr>
<tr>
<td>Sodium carbonate</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>1.0</td>
<td>2.0</td>
<td>3.8</td>
<td>5.0</td>
</tr>
<tr>
<td>1.0</td>
<td>3.6</td>
<td>6.7</td>
<td>7.0</td>
</tr>
<tr>
<td>1.0</td>
<td>3.7</td>
<td>5.4</td>
<td>6.3</td>
</tr>
<tr>
<td>Potassium fluoride</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>1.0</td>
<td>4.0</td>
<td>4.1</td>
<td>4.2</td>
</tr>
<tr>
<td>1.0</td>
<td>4.0</td>
<td>4.0</td>
<td>4.3</td>
</tr>
<tr>
<td>1.0</td>
<td>5.0</td>
<td>7.9</td>
<td>8.1</td>
</tr>
<tr>
<td>Sodium phosphate</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>1.0</td>
<td>4.0</td>
<td>3.6</td>
<td>5.6</td>
</tr>
<tr>
<td>1.0</td>
<td>4.7</td>
<td>4.7</td>
<td>4.8</td>
</tr>
<tr>
<td>1.0</td>
<td>4.0</td>
<td>4.5</td>
<td>4.6</td>
</tr>
<tr>
<td>1.0</td>
<td>3.1</td>
<td>4.0</td>
<td>4.3</td>
</tr>
<tr>
<td>Sodium sulfate</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>1.0</td>
<td>4.0</td>
<td>1.5</td>
<td>3.6</td>
</tr>
<tr>
<td>1.0</td>
<td>2.0</td>
<td>1.3</td>
<td>6.1</td>
</tr>
<tr>
<td>1.0</td>
<td>2.4</td>
<td>7.1</td>
<td>10.4</td>
</tr>
<tr>
<td>1.0</td>
<td>2.2</td>
<td>6.4</td>
<td>5.9</td>
</tr>
</tbody>
</table>
The slag-based binder was characterized by sufficiently high strength characteristics under normal conditions.

The strength after 28 days was 53.0–72.0 MPa, 7 days – 15.0–32.0 MPa. It should be noted that the samples of binders of this composition with the content of each activator of 4% or more after hardening under normal conditions had efflorescence.

In the presence of 1% sodium hydroxide and 0.5–2% soda, the activity of PSB increased to 72.0–90.0 MPa.

It was shown above that it is possible to obtain PSB with a sufficiently high activity under normal and curing conditions in the presence of small amounts of composite additives.

The composition of the composite additives includes alkali and sodium sulfate or soda. The first of these salts is active mainly under curing conditions, the action of the second one is more effective under normal conditions.

Since under normal conditions, the hardening process of binders is determined primarily by the content of soda, and not sodium sulfate, we made a synthetic soda-sulfate mixture (SSM) with a low content (30%).

In the slag samples, the soda content varies widely (25–60%).

Under normal conditions, with an alkali content of 1–2% and SSM of 0.5–2%, the strength of the samples reached 28.0–36.0 MPa after 3 days, 38.0–54.0 MPa after 7 days, and 60.0–80.0 MPa after 28 days.

The lowest strength was noted for the composition with the content of each activator 2%. The simultaneous presence of even small amounts (0.5%) of potassium fluoride and sodium hydroxide dramatically increases the strength of the samples. After 7 and 28 days, it is 36.0 and 60.0 MPa, respectively. When the amount of salt increases to 4%, the strength does not change significantly.

The maximum strength parameters were observed at 1% sodium hydroxide and 0.5–2% potassium fluoride. The binders of the considered composition are characterized by high strength of 80.0–90.0 MPa under curing conditions, which is obtained at optimal activator contents (0.5–1% sodium hydroxide and 1–4% potassium fluoride).

A mixture of sodium hydroxide and sodium phosphate was used to activate PSB in curing conditions.

A composite additive consisting of 1% sodium hydroxide and 0.5–4% sodium phosphate provides the strength of samples at 3 days age in the range of 32.0–41.0 MPa, at 7 days age – 39.0–66.0 MPa, at 28 days age – 57.5–68.0.

The composition of cement with soda turned out to be more effective in its effect on the early hardening of phosphorus slag binders. In this case, the addition of the same amounts of activators made it possible to obtain fast-hardening binders with a strength of 35.0–50.0, 41.0–83.0 and 51.0–78.0 MPa after 3, 7 and 28 days, respectively. With a constant cement content, the strength increases with an increase in the amount of soda in the range of 1–4%. The effect of cement with a constant content of the salt component is less pronounced.

A similar relationship takes place in the curing conditions. The maximum strength values (72.0–84.0 MPa) were observed at the content of 2–4% cement and soda. The strength of the samples based on different slags did not differ significantly. The mechanisms of hydration of composite binders containing phosphorus slag at different temperatures are given in the work [20].

Binders containing a composition of cement with a synthetic soda-sulfate mixture showed slightly lower strength after curing. With a constant cement content (2–4%), the strength indicators increase with an increase in the proportion of the salt additive, reaching its maximum value at 4% (60.0–65.0 MPa).

A similar dependence of strength on the content of activators is observed for binders that harden under normal conditions.

At the same time, the strength of the samples hardening during 3 and 7 days reaches 30.0–45.0 and 60.0 MPa, respectively.

The strength parameters of the samples with the addition of salt and water after 3 days of storage do not differ significantly.

The strength of binders after 28 days, as well as after curing, is significantly lower compared to binders containing soda.

Binders activated with 4% cement and 1–2% potassium fluoride had a strength of 58.0 and 72.0 MPa, respectively, after curing (Table 5), which is 1.8–2.8 times higher than the strength of samples with the addition of only the first activator.

After 7 and 28 days, it is 25.0–37.0 MPa and 46.0–70.0 MPa, respectively. When the amount of salt increases to 4%, the strength does not change significantly.

The maximum strength values were observed at 2–4% cement and 2–4% potassium fluoride.

### Table 5

<table>
<thead>
<tr>
<th>Amount of activators, wt.%</th>
<th>Binding strength, MPa</th>
<th>Compressive strength, MPa</th>
</tr>
</thead>
<tbody>
<tr>
<td>Portland cement</td>
<td>Salt</td>
<td>Normal conditions, days</td>
</tr>
<tr>
<td>Sulfate carbonate</td>
<td>4.0</td>
<td>4.0</td>
</tr>
<tr>
<td></td>
<td>2.0</td>
<td>2.0</td>
</tr>
<tr>
<td></td>
<td>1.0</td>
<td>1.0</td>
</tr>
<tr>
<td></td>
<td>0.5</td>
<td>0.5</td>
</tr>
<tr>
<td>Potassium fluoride</td>
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<td>Sodium sulfate</td>
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The binders of the composition under consideration are characterized by high strength both under curing and normal conditions, which, at optimal activator contents (4%...
cement and 2–4 % potassium fluoride), reaches 72.0–80.0 MPa under curing and 70.0–75.0 MPa under normal conditions, respectively.

A composition of cement with sodium phosphate was also used to activate the phosphorus slag binders under curing and normal conditions [21].

The strength characteristics of the binders of the “slag-cement-sodium phosphate” system are determined by the quantitative ratio of activators.

Small amounts of salt 0.5–1 % do not cause hardening of the binders at 2–4 % cement content when stored for 7 days.

After 28 days, these samples had a strength of 16.0–23.0 and 38.0–50.0 MPa, respectively. A higher phosphate content (2 %) with 4 % cement addition contributes to a significant increase in strength up to 62.0 MPa.

In the curing conditions, a similar pattern is observed: an increase in the amount of salt from 0.5 to 4 % causes an increase in strength from 54.0 to 75.0 MPa with 4 % cement addition. The binding compositions “Portland cement-sodium hydroxide” are characterized by sufficiently high strength indicators and in normal conditions their strength after 28 days was 30.0–58.0 MPa. It is important to note a more beneficial effect of curing in comparison with normal conditions on this system. The strength of the samples after curing was 34.0–68.0 MPa.

As can be seen from Table 6, the addition of alkali increases the pH and, the alkaline balance of sodium sulfate increases to a greater extent than that of sodium carbonate. When adding the compositions (NaOH+salt) to the slag, the pH of the silicate system decreases significantly after 48 hours.

Knowing the limits of the existence of various forms of orthosilicic acid and its ions according to the ideas [22], the activity of binders is higher at a lower salt content under curing conditions (0.5 %) than under normal conditions (2 %). With the addition of 1.0 % alkali and 0.5 % salt, the strength of the samples after curing (MPa) was: (98) Na₂CO₃>(98) KF>(86) Na₃PO₄>(76) Na₂SO₄, and under normal conditions with the addition of 1.0 % alkali and 2 % salt, the strength of the samples was: (90) Na₂CO₃>(80) KF>(72) Na₂SO₄>(68) Na₃PO₄.

A number of strength parameters correspond to a number of values of the dissociation constants of the acids forming the corresponding solid bases in the system: KH₂CO₃ (4.4×10⁻³)<KHF (6.6×10⁻⁴)<K₃PO₄ (7.5×10⁻⁵)<K₂SO₄(2×10⁻⁶) [22, 23].

The highest degree of silicic acid polymerization in the “slag-NaOH-Na₂CO₃-H₂O” system, expressed by the formation of a large number of different forms of orthosilicic acid in comparison with other systems considered, explains its first position in the strength series.

The formation of only two forms of orthosilicic acid in the “slag-NaOH+Na₂SO₄-H₂O” system explains the lowest activity of the obtained binder.

Probably, the hardly soluble compounds formed act on the principle of acid-base catalysts [24] the pH of the medium of the alkaline systems, which, in turn, affects the degree of polymerization, accelerates the main reactions of hydration of silicates.

5. Research results of the influence of the composition of activators on phosphorus slag binders

5.1. Selection of the composition of small amounts of sodium hydroxide with pure alkali metal salts

The selected and used pure salts of sulfate, sulfite, carbonate, sodium phosphate, potassium fluoride by anion formed poorly soluble compounds: CaCO₃, Ca₂(PO₄)₂, CaSO₄.

The effect of the combined presence of alkali and salt is especially noticeable for small amounts of sodium hydroxide. It was found that in an alkaline medium, the activity of binders is higher at a lower salt content under curing conditions (0.5 %) than under normal conditions (2 %). With the addition of 1.0 % alkali and 0.5 % salt, the strength of the samples after curing (MPa) was: (98) Na₂CO₃>(98) KF>(86) Na₃PO₄>(76) Na₂SO₄, and under normal conditions with the addition of 1.0 % alkali and 2 % salt, the strength of the samples was: (90) Na₂CO₃>(80) KF>(72) Na₂SO₄>(68) Na₃PO₄.

5.2. Selection of the composition of small amounts of sodium hydroxide with waste containing the corresponding compounds

Selected and used soda-sulfate waste from caprolactam production – soda-sulphate mixture – SSM containing 30 % soda and 70 % sodium sulfate, by-products of superphosphate plants containing low sodium fluoride, sodium fluoride and sodium phosphate by anion formed difficult-to-dissolve compounds: CaCO₃, Ca₂(PO₄)₂, CaSO₄.

A similar dependence of strength on the content of activator is observed for binders containing waste. The hydration and strength of the binder depend on the nature of the anion that forms poorly soluble compounds. The strength in the obtained systems changes in the series (MPA): CaCO₃(60)>CaF₂(50)>Ca₃(PO₄)₂(50)>CaSO₄(30).

5.3. Study of the effect of sodium hydroxide with pure alkali metal salts under normal conditions

Low-alkaline solutions on phosphorus slag binders are characterized by a slow set of strength in the early stages
of hardening, especially for aged slags. The strength with a sodium hydroxide content of 1–4 % after curing at 95 °C according to the 2+7+2 h mode for slag specimens of various batches was in the range of 50.0–70.0 MPa. Samples of binders of normal hardening at the age of 28 days with a sodium hydroxide content of 0.5; 1.0; 2 and 4 % had the strength of 20.3; 35.4; 45.6; 55.8 MPa, respectively.

With the addition of 1.0 % alkali and 0.5 % salt, the strength of the samples after curing (MPa) was: (98) Na₂CO₃>(98) Na₂CO₃>(98) KF>(86) Na₃PO₄>(76) Na₂SO₄, and under normal conditions with the addition of 1.0 % alkali and 2 % salt, the strength of the samples was, (MPa): (98) Na₂CO₃>(98) KF>(86) Na₃PO₄>(76) Na₂SO₄.

5. Study of the effect of cement with pure alkali metal salts under normal conditions

Binders containing a composition of cement with salts under normal conditions and after curing showed a slightly lower strength than in an alkaline medium. At a constant cement content (4 %), the strength indicators increase with an increase in the proportion of the salt additive, reaching at 4 % its maximum value (MPa): (98) Na₂CO₃>(98) KF>(86) Na₃PO₄>(76) Na₂SO₄ under normal conditions.

6. Discussion of experimental results of the production technology of phosphorus slag binders

The research results according to all the assigned tasks from the problem area are clearly shown in section 5, in the text of the paper, in Tables 4, 5. This work shows us the proposed way to solve the problem, an alternative way is the choice for our colleagues.

The technology for the production of phosphorus slag binders is based on the theoretical prerequisites for the destruction of silicon and calcium-oxygen polyhedral groups of slag under the influence of various activating components.

The use of alkali metal salts, the anions of which form poorly soluble compounds with calcium or phosphorus slag in a composition with a low content of sodium hydroxide, which has a pronounced activating ability and reduces the degree of hydrolysis of sodium metasilicate, makes it possible to achieve high strength parameters and setting times required for their industrial use.

The addition of chemically pure sodium hydroxide as an alkaline component to the system limits its use on a construction scale.

The inconsistency of the existing regulatory documents that allow obtaining binders without cleaning and non-permissive ones requires additional research in different countries to develop a unified approach.

In connection with the tightening of requirements for environmental and sanitary standards, the absence of a database on waste that could be used as activating components.

7. Conclusions

1. The hydration and strength of the binder depend on the nature of the anion that forms poorly soluble compounds. The strength in the obtained systems changes in the series (MPa): CaCO₃ (70)>CaF₂ (60)>Ca₃(PO₄)₂ (50)>CaSO₄ (30).

2. The effect of the nature of activators on pH was determined. The content of individual forms of orthosilicic acid is assumed depending on the pH value in the systems under study.

3. A number of strength indicators correspond to a number of values of the dissociation constants of acids forming the corresponding solid bases in the system: KH₂CO₃ (4.4×10⁻³)<KHF (6.6×10⁻¹)<KH₃PO₄ (7.5×10⁻³)<KH₂SO₄ (2×10⁻¹). Thus, alkali metal salts, the anions of which form poorly soluble compounds with calcium, when used together with a small amount of alkali, make it possible to obtain highly active alkali-alkali binders that harden under normal and curing conditions.

4. The data obtained indicate the advantages of using phosphorus slag binders and various industrial waste with a low content of alkaline compounds in the production. Such wastes are alkaline solutions of coke-chemical industries, solutions obtained during air purification at metallurgical plants, waste products from β-naphthol production, soda sulfate waste from caprolactam production, waste soda production, by-products of superphosphate plants.

References


