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UDC 796.015:613.2 DOI: 10.15587/2312-8372.2019.187673

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RESEARCH OF OXIDATIVE STABILITY OF VEGETABLE OILS FOR USE IN SPORT NUTRITION

Об'єктом дослідження є соняшникова, соняшникова високоолеїнова та ріпакова олії, які є важливими у виробництві продуктів харчування для людей з підвищеними фізичними навантаженнями. Такі продукти мають бути збалансовані за компонентним складом та збагачені біологічно активними речовинами, вітамінами, антиоксидантами. Олії та жири використовують як важливий компонент спортивного харчування. Розповсюдженою олією є соняшникова олія, яка забезпечує раціон лінолевою кислотою, а також ріпакова олія, яка збагачує продукти ліноленовою кислотою. Одним з найбільш проблемних місць використання олій є окиснювальне псування, оскільки це призводить до утворення небезпечних для здоров'я сполук. В ході дослідження використовувалися метод активного кисню та прилад Oxitest (Італія). Перший метод дозволяє безпосередньо оцінювати вміст продуктів окиснення в олії через значення пероксидного числа, другий метод дозволяє вимірювати витрату кисню під час процесів окиснення в олії.

Проведено дослідження окиснювальної стабільності олії соняшникової методом активного кисню та методом окиснення з використанням приладу Oxitest за температури 110 °C. Встановлено, що періоди індукції, визначені двома методами, є близькими (2 год. 40 хв. та 2 год. 43 хв., відповідно). Встановлено закономірність зниження періоду індукції олії соняшникової під час підвищення температури методом окиснення на приладі Oxitest. Використано температури 90, 100 та 110 °C. Виконано порівняльне дослідження окиснювальної стабільності соняшникової, соняшникової високоолеїнової та ріпакової олій. На підставі одержаних даних наведено порівняльну характеристику стійкості дослідних олій до окиснення та надано рекомендації для їх використання у спортивному харчуванні. Одержані дані дозволять оцінювати вплив кожної з дослідних олій на строк придатності олієвмісного продукту.

Ключові слова: спортивне харчування, олія соняшникова, олія ріпакова, окиснювальна стабільність, період індукції.

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1. Introduction

Current nutritional requirements require the consumption of polyunsaturated fatty acids, essential amino acids and minerals in sufficient quantities. But usually modern nutrition does not fully satisfy the need for such important components [1, 2]. A particularly acute issue of good nutrition is relatively certain categories of the population, in particular, people who engage in sports activities. Preparation for the competition, daily training, high psychological endurance require special nutrition, which will contribute to high endurance and quick recovery [3–5]. In order to meet the needs of the body of a person involved in sports, special products are currently being developed, for example, highprotein products, confectionery, beverages, etc. [6-8]. At the same time, one of the most important components of such products is oils and fats, which provide the necessary balance of fatty acids valuable for metabolism.

Among used oils, sunflower and corn are a valuable source of linoleic acid (9, 12 octadecadienoic, group ω -6); soybean, rapeseed, flaxseed and cameline α -linolenic (group ω -3). These acids are not synthesized in the human body and can only get into it as part of food products. They take part in the synthesis of hormones and in the metabolism in cells, in the creation of cell membranes. So, the study of these oils from the point of view of their use for creating sports nutrition is of significant interest [9].

The high oleic sunflower oil as a component of many types of food products is also attracting more and more attention. High oleic oils and oils are characterized by a high content of monounsaturated fatty acids. This is an advantage in terms of food hygiene, since it means a lower content of saturated acids, the absence of trans isomers of fatty acids, and also provides increased resistance to oxidation. The latter property is undeniably important in the production and use of all fat-containing products due to the tendency of oils and fats to oxidative damage [10, 11].

Oxidative processes lead to the formation of substances harmful to the human body, which leads to an increasing interest of researchers in studying the laws of oxidation of oils and fats with various fatty acid compositions and the search for new effective and environmental antioxidants [12].

Various methods are used to determine oxidative stability, among which the barometric method for determining the oxidation stability of oils, fats, and fat-containing products using the Oxitest device (Italy) is distinguished. In this case, increased temperature and pressure are used. This method reliably allows to determine the resistance of the sample to the action of oxygen, which allows to draw conclusions about the freshness, effectiveness of antioxidants and shelf life of products [12].

Therefore, it is relevant to study the oxidative stability of vegetable oils, which are promising components of various types of food products, including sports.

Thus, *the object of research* is sunflower, high oleic sunflower and rapeseed oil, which are important in the production of food for people with increased physical activity.

And *the aim of research* is a comparative study of the oxidative stability of vegetable oils (sunflower, high oleic sunflower and rapeseed) by various methods (active oxygen method and using the Oxitest device).

2. Methods of research

2.1. The investigated materials and equipment used in the experiment. The following reagents and materials were used in this study:

 refined deodorized rapeseed oil in accordance with DSTU 8175:2015;

- high oleic refined deodorized sunflower oil in accordance with current regulatory documents;

refined deodorized sunflower oil in accordance with DSTU 4492:2017;

 $-\,$ gas cylinder with oxygen purity not lower than 99.95 % according to the current regulatory documentation;

rectified ethyl alcohol according to DSTU 4221:2003;

- distilled water current regulatory documentation;

 potassium iodide qualification «especially pure» according to the current regulatory documentation;

– soluble starch according to the current regulatory documentation;

 sodium thiosulfate according to the current regulatory documentation;

acetic acid according to the current regulatory documentation;

chloroform according to the current regulatory documentation;

- Oxitest device (Italy).

2.2. Method for determining the oxidative stability of sunflower oil by the method of active oxygen. The oxidative stability of sunflower oil samples is determined by the accelerated method of active oxygen, which provides for the determination of the induction period for the change in the peroxide number during oil exposure at a given temperature. 10 g of an oil sample was introduced into the glass containers, placed in an oven heated to a predetermined temperature. To determine the peroxide value from the oven at regular intervals, the corresponding bottle was selected. The value of the induction period of sunflower oil was determined graphically from the growth curves of peroxide numbers by the tangent method.

The peroxide value is determined by the standard method according to DSTU 4350:2004.

2.3. Method for determination of oxidative stability of oil samples using the Oxitest instrument. The Oxitest device (Fig. 1) allows to assess the resistance degree of various types of products (oils, fats, flour, cheeses, milk powder, meat products, sauces, seeds and other fat-containing products) to oxidation. In this case, the change in pressure is controlled due to oxygen absorption during the oxidation of the prototype in two autonomous thermostatic chambers.

The chambers are filled with a prototype and carefully closed, after which heating of the chamber to a predetermined temperature begins, then oxygen is supplied to the chamber. After the set pressure is set to 6 bar (0.6 MPa), the oxygen supply is pricked up and the reaction time measurement is turned on.



Fig. 1. Oxitest device

The study of oil oxidation is carried out according to the nature of the curve, which is automatically based on the graphical dependence of pressure on reaction time. An example of a graphical dependence is shown in Fig. 2.

At the end of the test, the user can use two equivalent methods for calculating the induction period: the graphical method and the least squares method.

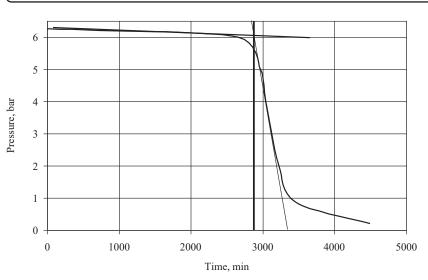


Fig. 2. Graphical dependence of pressure on reaction time for calculating the induction period

The graphical method allows to automatically build two tangent lines – the pressure curve before and after bending and get the point of intersection. From this intersection point, a perpendicular to the abscissa is drawn. The numerical value on the abscissa axis, on

which the perpendicular is dropped, is the numerical value of the induction period.

CHEMICAL ENGINEERING:

In the case of the least squares calculation, the graphical dependence first displays four vertical lines that define two groups of curve points, are reduced to two lines. The intersection point of these two lines gives the value of the induction period of the sample.

In the work, the induction period is calculated using the graphical method. The error in calculating the experimental data obtained using the Oxitest instrument does not exceed 3 % [12].

3. Research results and discussion

For a comparative analysis of the determination of the induction period by various methods, the active oxygen method and the Oxitest device were used. Refined deodorized sunflower oil was used as research oil. The study temperature was 110 °C.

Fig. 3 shows the dependence of the oxidation time on the oxygen pressure in the thermostatic chamber of the Oxitest device.

As can be seen from Fig. 3, the oxygen pressure in the chamber for a certain time decreases rather slowly, which corresponds to the diffusion of oxygen from the contact surface of the oil with oxygen. Thus, in this period of time, the oil remains stable until oxidative processes. Then there comes a moment of a sharp change in pressure, when oxygen begins to be actively spent on oil oxidation processes, while the pressure in the chamber decreases sharply. After the end of this process, the pressure in the chamber remains constant and does not exceed 1.58 bar (0.158 MPa). When processing the graphic dependence, the induction period is obtained – 2 hours 43 minutes.

In parallel, a study is conducted to determine the period of induction of sunflower oil by the method of active oxygen, after changing the values of the peroxide value under conditions of oil soaking at a temperature of 110 °C.

The results of determining the induction period by the method of active oxygen are shown in Fig. 4, as the dependence of the peroxide number on the oxidation time.

When determining the induction period by the method of active oxygen, pure oxygen and additional pressure are not used. When processing the obtained dependence (Fig. 4), an induction period of 2 hours and 40 minutes is obtained.

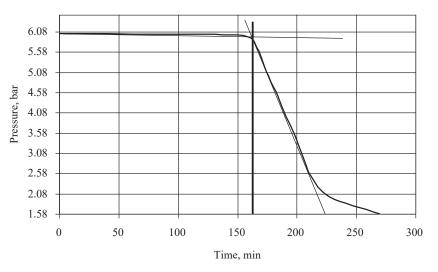
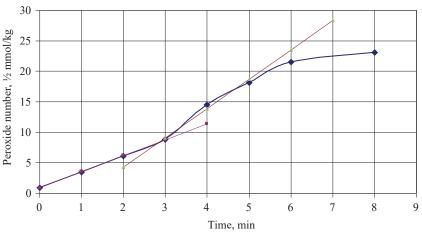
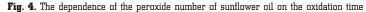


Fig. 3. The dependence of the oxidation time on the oxygen pressure in a thermostatic chamber





Thus, the oxidative stability of sunflower oil at a temperature of 110 °C is determined by two different methods. Despite the difference in the principles of the methods, the values of the induction periods are determined to be close. The method of active oxygen allows to determine the induction period through a direct indicator of oxidative damage - the peroxide number, which is reliably characterizes the accumulation of peroxide compounds in the prototype. But this method is time-consuming, requires the use of special reagents and materials, as well as high precision laboratory tests. The Oxitest method is an innovative modern tool for studying the oxidative stability of various foods. It does not require high operator qualifications, additional materials and allows to reliably determine the induction period through the oxygen consumption rate of the prototype. So, the Oxitest device has undeniable advantages over the active oxygen method, since it automatically determines the induction period and reduces the influence of the human factor on the final result.

A study was made of the effect of temperature on the induction period of refined deodorized sunflower oil on an Oxitest device at the temperatures of 90, 100 and 110 °C. The duration of the study with a decrease in temperature increases from 5 hours and 60 minutes at a temperature of 110 °C until 22 hours 50 minutes at a temperature of 90 °C. The induction period of sunflower oil with different temperatures is shown in Fig. 5.

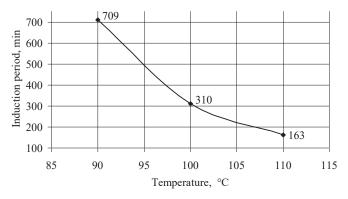


Fig. 5. The induction period of sunflower oil with different temperatures (709 minutes equal to 11 hours 49 minutes, 310 minutes equal to 5 hours 10 minutes, 163 minutes equal to 2 hours 43 minutes)

As can be seen from the presented data (Fig. 5), with an increase in temperature from 90 to 100 °C, the induction period decreases by 2.3 times, with an increase in temperature from 100 to 110 °C, the induction period decreases by 1.9 times. Thus, a temperature increase of 10 °C reduces the induction period by about half.

For a quick assessment of the oxidative stability of a sample, it is advisable to use a temperature of 100 $^{\circ}\mathrm{C}$ and higher.

The oxidative stability of sunflower, high oleic sunflower and rapeseed oils using the Oxitest device are compared at a temperature of 100 °C. Table 1 shows the results obtained for periods of induction of research oils.

As can be seen from the data in Table 1, behind the Oxitest device, the smallest induction period has sunflower oil, and the highest sunflower oil is high oleic. Rapeseed oil has an induction period of 1.8 times more than sunflower oil.

The values of the induction periods of research oils

Sample name	Induction periods
Sunflower oil	5 hours 10 minutes
High oleic sunflower oil	11 hours 50 minutes
Rapeseed oil	9 hours 23 minutes

Based on the data obtained on the study of the oxidative stability of types of oils that are important for use in sports practice, it is found that the highest oxidative stability is in the case of high oleic sunflower oil. This is due to the increased content of monounsaturated acids in comparison with other experimental samples of oils. The obtained induction periods allow to evaluate the oxidative behavior of oils in the composition of food products and make decisions on the component composition of oils in the product.

It should be noted that the method of the Oxitest device relative to the method of active oxygen makes it possible to work not only with fats and oils in pure form, but with fat-containing products, without isolating the fat phase, reduces the complexity of the analysis. In addition, this eliminates the distortion of the results for additional operations with the prototype, which can lead to a change in the oxidative state. By extracting fat in products such as seeds and nuts, not all natural inhibitors

found in fruits and seeds can be removed, and thus the result of determining the induction period can also be distorted. Cosmetics and food products that contain water and fatty components, such as cheeses, cosmetic creams, lipsticks and other products, can also be used in their original form, without removing fat components from them. In such cases, the experiment at elevated pressure in the measuring chamber of the device will ensure the stability of the water contained in the object (preserving it in the liquid phase), even at temperatures above 100 °C.

This makes it possible, in addition to pre-setting the oxidative stability of oils, to further determine the oxidative stability of the finished product for sports purposes. The data obtained using the Oxitest device makes it possible to reliably and efficiently select the optimal composition of a multi-component product in order to ensure the required shelf life.

4. Conclusions

A comparative analysis of the induction period of sunflower oil is made by the method of active oxygen and on the device Oxitest. It is found that the values of the induction periods are close (2 hours 40 minutes and 2 hours 43 minutes). But at the same time, the Oxitest device has several advantages along with the active oxygen method.

Using the Oxitest instrument, the regularity of the change in the oxidative stability of sunflower oil with temperature is studied. With an increase in temperature for every 10 °C, the period of oil induction is almost halved.

The oxidative stability of sunflower, high oleic sunflower, rapeseed oils are compared using the Oxitest method. The periods of oil induction are determined, among which the highest oxidative stability is found in high oleic sunflower oil. Rapeseed oil has an induction period of 1.8 times more than sunflower oil.

The obtained results are important in the production of balanced multi-component products for sports nutrition, since the component determines the shelf life of the product, mainly oils and fats. Moreover, they are the most important functional part of many types of sport products.

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