

**IMPROVEMENT OF PRODUCTION BY AEROSOL NANOCATALYSIS TECHNOLOGY WITH MECHANICAL ACTIVATION OF CATALYST PARTICLES**

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Currently, the chemical and oil industry uses a heterogeneous catalysis with catalyst on support. The results of a large number of studies marked its main drawbacks. They are insufficient strength and heat resistance of supported catalysts. The solution to this problem is to increase the strength of the catalyst support. This article proposes a new future-proof solution. It is the use of the aerosol nanocatalysis technology for the processes of chemical and oil industries. The basic principle of this technology is the total exclusion of the use of the support from industry practices. This principle together with the chemical and mechanical activation of the catalyst showed an increase in the reaction rate and reduction of the catalyst concentration. It also influences the contact time reduction and productivity increase in the main component. Mechanical activation of the catalyst particles by the dispersing material successfully multiply return the catalyst into the reactor. The aerosol nanocatalysis technology has no analogues in the world. It will look at the industrial world in a new way. It is planned in the future to carry out targeted synthesis of chemical processes in the conditions of an aerosol nanocatalysis technology.

Keywords: aerosol nanocatalysis, mechanical activation, heterogeneous catalyst, activity, strength and heat resistance of surface.

References

- Satterfield, C. (1980). *Heterogeneous Catalysis in Practice*. McGraw-Hill, 416.
- Krilov, O. V. (2004). *Geterogennyi kataliz*. Moscow: Akademiya, 679.
- Glikin, M. A. (2000). *Effektivnost' i vzryvobezopasnost' protsessov himicheskoi tekhnologii*. Kyiv: Print Ekspres, 390.
- Parmon, V. N. (1997). *Akademik Georgii Konstantinovich Boreskov. Ocherki, materialy, vospominaniia*. Novosibirsk: IK SO RAN, 460.
- Glikin, M. A. (1980). Vliianie himicheskikh reaktzii na vzryvopasnost' tekhnologicheskoi sredy. *Teoreticheskie osnovy himicheskoi tekhnologii*, 14 (5), 770–772.
- Boreskov, G. K. (1986). *Geterogennyi kataliz*. Moscow: Nauka, 304.
- In: Anderson, R. B. (1968). *Experimental Methods in Catalytic Research*. New York and London: Academic Press, 498. doi:10.1016/c2013-0-06162-2
- Kiperman, S. L. (1991). Kinetic models of heterogeneous catalytic reactions (review). *Bulletin of the Academy of Sciences of the USSR Division of Chemical Science*, 40 (12), 2350–2365. doi:10.1007/bf00959702
- Glikin, M. A. (1996). Aerosol'nyi kataliz. *Teoreticheskie osnovy himicheskoi tekhnologii*, 30 (4), 430–435.
- Glikin, M. A., Pihovnikov, B. I., Novitskii, V. S., Memedliayev, Z. N., Kutakova, D. A., Viks, I. N., Prin, E. M.; assignee: Limited Liability Company «Nitrohim». (20.06.1997). Sposob osushchestvleniia gazofaznykh himicheskikh protsessov (aerosol'nyi kataliz). *Patent 2081695 RU, MKI V 01 J 8/08, 8/32*. Appl. № 94011388. Filed 01.04.1994. Bull. № 17, 7.
- Boreskov, G. K., Levitskii, E. A., Ismagilov, Z. R. (1984). Szhiganie topliv i kataliticheskie generatory tepla. *Rossiiskii himicheskii zhurnal*, XXIX (4), 19–25.
- Ismagilov, Z. R. (1985). Catalytic heat generators for technological processes. *Kagaki to Kogyo*, 38 (3), 262.
- Levitskii, E. Ya., Glikin, M. A., Kutakova, D. A. et al. (1981). Likvidatsiia vrednykh veshchestv v gazovykh vybrosakh pri kataliticheskom szhigani stochnykh vod. *Tez. dokl. III Vsesoiuznoi konf. «Kataliticheskoe okislenie gazov»*. Novosibirsk, 157–160.
- Boreskov, G. K., Matros, Y. S. (1983). Unsteady-State Performance of Heterogeneous Catalytic Reactions. *Catalysis Reviews*, 25 (4), 551–590. doi:10.1080/01614948308078056
- Ismagilov, Z. R., Arendarskii, D. A., Kirichenko, O. A., Baranik, G. B., Moroz, E. M., Ushakov, V. A. (1989). Issledovanie reaktzii i katalizatorov szhiganiia topliv. IV. Genezis fazovogo sostava nanesennykh oksidnykh aliumomednohromovykh katalizatorov. *Kinetika i kataliz*, 30 (4), 918–926.
- Moroz, E. M., Ushakov, V. A., Kosmambetova, G. R., Shepelin, A. P., Levitskii, E. A. (1988). Dispersnost' i sostav termoo-brabotannykh aliumoplatinovykh katalizatorov. *Kinetika i Kataliz*, 29 (6), 1446–1451.
- Parmon, V. N., Simonov, A. D., Sadykov, V. A., Tihov, S. F. (2015). Kataliticheskoe szhiganie: dostizheniia i problemy. *Fizika goreniia i vzryva*, 51 (2), 5–13.
- Noskov, A. S., Parmon, V. N. (2008). Kataliticheskie tekhnologii dlia rasshireniia toplivno-syr'evoi bazy Rossii za schet netradsionnykh istochnikov uglerodsoderzhashchego syr'ia. *Gazohimiia*, 2 (1), 20–24.
- Tiul'pinov, A. D. (1986). *Predel'nye uslovia gomogenного goreniia v kataliticheskikh generatorah tepla*. Moscow, 18.
- Kutakova, D. A. (1987). *Glubokoe okislenie promyshlennykh ot-hodov v kataliticheskikh generatorah tepla*. Moscow, 172.
- Ermakov, Yu. I., Zaharov, V. A., Kuznetsov, B. N. (1980). *Zakreplennyye kompleksy na oksidnykh nositeliah v katalize*. Novosibirsk: SO Nauka, 248.
- Khimach, N. Yu., Polunkin, Ye. V. (2012). Nanostrukturovani katalizatory. *Kataliz i neftehimiiia*, 21, 86–98.
- Patriliak, K. I., Patriliak, L. K. (2012). Teoriia J. A. Ola i geterogennyi kataliz. *Kataliz i neftehimiiia*, 20, 6–21.
- Glikin, M., Kutakova, D., Prin, E. (1999). Unsteady processes and aerosol catalysis. *Chemical Engineering Science*, 54 (20), 4337–4342. doi:10.1016/s0009-2509(99)00133-5
- Frank-Kamenetskii, D. A. (1967). *Diffuziia i teploperedacha v himicheskoi kinetike*. Moscow: Nauka, 502.
- Glikina, I. M. (2005). *Osnovy tekhnologii aerosolnoi nanokatalitichnoi pererobky orhanichnykh spolk u vibrozridzhenomu shari*. Lviv, 162.
- Kudryavtsev, S. A. (2006). *Osnovy tekhnologii polucheniia benzinovoi fraktsii i etilena aerosol'nykh nanokatalizom*. Lviv, 154.
- Riazantsev, O. I., Kardashuk, V. S. (2011). Metody ta programno-tekhnichni zasoby avtomatyzatsii keruvanniia protsesom aerosolnoho nanokatalizu. *Radioelektronika, informatyka, upravlinnia*, 1, 164–171.
- Glikin, M. A., Kutakova, D. A., Prin, E. M., Glikina, I. M., Volga, A. I. (2000). Geterogennyi kataliz na poristoii strukture i v aerazole. *Kataliz i neftehimiiia*, 5–6, 92–101.
- Prin, E. M. (2002). *Aerosol'nyi kataliz i ego primenenie v okislenii ammiaka i obezvezhivaniu soedinenii sviazannogo azota*. Kharkiv, 156.
- Kascheev, A. S., Glikina, I. M., Kudryavtsev, S. A., Mamedov, B. B. (2012). Issledovanie aktivnosti Si/Zr katalizatora v protsesse krekinga vakuumnogo gazoilia aerosol'nykh nanokatalizom. *Voprosy himii i himicheskoi tekhnologii*, 1, 85–89.

INVESTIGATION OF BARRIER PROPERTIES OF PACKAGING POLYMER FILM MATERIALS

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The main types of interactions of packaged food product and packaging are analyzed. These interactions may occur between the polymer film packaging material and the product. It is found that the most simple to implement an insulation of internal space of polymer film packaging from environmental impact. It is noted that the permeability of gases and gas mixtures through nonporous film polymer packaging is typically considered within the «dissolution – diffusion» mechanism. Usually the study of these processes requires special equipment.

The main factors that affect the permeability of the polymer film packaging material are analyzed. It is noted that the permeability of the multilayer polymeric film materials is dependent on the physical and chemical properties of the components and process parameters. The main parameters are: temperature, pressure

and concentration of the substance. It is noted that molecular interactions with packaging materials can be classified into the following types: permeability of compounds through the polymer film packaging; sorption of compounds by polymer film packaging and migration of compounds from a polymer film packaging. It is indicated that obtained results may be used for the design of packaging film materials with the necessary and stable barrier properties.

The technique of experimental studies of the barrier properties of various polymer film packaging materials and absorption capacity of packaged products is proposed and its results are described. This technique is realized using available equipment. The dynamics of moisture penetration inside the packaging is investigated. It is found that the absorption capacity of silica gel remained stable for 15–20 days from the beginning of the experiment. The highest moisture permeability has an ethylene and vinyl acetate copolymer and low-density polyethylene, and the lowest – polypropylene

Keywords: polymer packaging, moisture permeability, diffusion, barrier properties, mass transfer.

References

- Selke, S. E. M., Culter, J. D. (2015). *Plastics Packaging: Properties, Processing, Applications, and Regulations*. Munich: Carl Hanser Verlag, 487. doi:10.3139/9783446437197
- Hanlon, J., Kelsey, R., Forcinio, H. (1998). *Handbook of Package Engineering, Third Edition*. CRC Press, 698. doi:10.1201/b18045
- Muravin, Ya. G., Tolmacheva, M. N., Dodonov, A. M. (1985). *Primenenie polimernykh i kombinirovannykh materialov dlia upakovaniia pishchevykh produktov [The using of polymeric and composite materials for packaging of foodstuffs]*. Moscow: Agropromizdat, 205.
- Steele, R. (2004). *Understanding and Measuring the Shelf-Life of Food*. Elsevier, 448. doi:10.1016/b978-1-85573-732-7.50002-0
- Kryzhanovskii, V. K., Kerber, M. L., Burlov, V. V., Panimatchenko, A. D. (2004). *Proizvodstvo izdelii iz polimernykh materialov [Manufacture of products from polymeric materials]*. Sankt-Petersburg: Professiia, 464.
- In: Gul', V. E. (1976). *Polimernye plenochnye materialy [Polymeric film materials]*. Moscow: Himiia, 248.
- Golub, O. V., Vasil'eva, S. B. (2005). *Upakovka i hranenie pishchevykh produktov [Packaging and storage of food products]*. Kemerovo: KTIPP, 148.
- Samoilov, V. Ya., Ostapchuk, V. V., Taranenko, I. M. (2012). *Tekhnologii i oborudovanie upakovochnykh materialov [Technology and equipment for packaging materials]*. Kharkiv: NAU n. a. N. E. Zhukovskogo «KhAI», 71.
- Efremov, N. F., Kolesnichenko, M. G. (2011). *Tekhnologii upakovochnogo proizvodstva [Technology of the packaging production]*. Moscow: MGUP, 350.
- Jovanovic, S., Dzunuzovic, J. (2011). Development directions of packaging made from polymer materials. *Hemijaska Industrija*, 65 (6), 621–635. doi:10.2298/hemind111018087j
- Siracusa, V. (2012). Food Packaging Permeability Behaviour: A Report. *International Journal of Polymer Science*, 2012, 1–11. doi:10.1155/2012/302029
- Suput, D., Lazić, V., Levic, L., Krkić, N., Tomović, V., Pezo, L. (2013). Characteristics of meat packaging materials and their environmental suitability assessment. *Hemijaska Industrija*, 67 (4), 615–620. doi:10.2298/hemind120907104s
- Atmani, R., El Kouali, M., Talbi, M., El Brouzi, A., Ouzidan, F. (2015). Study of the Profile Concentration of a Chemical Through a Sandwich Packaging. *Oriental Journal of Chemistry*, 31 (3), 1659–1662. doi:10.13005/ojc/310344
- Schmid, M., Dallmann, K., Bugnicourt, E., Cordon, D., Wild, F., Lazzeri, A., Noller, K. (2012). Properties of Whey-Protein-Coated Films and Laminates as Novel Recyclable Food Packaging Materials with Excellent Barrier Properties. *International Journal of Polymer Science*, 2012, 1–7. doi:10.1155/2012/562381
- Schmid, M., Sangerlaub, S., Miesbauer, O., Jost, V., Werthan, J., Stinga, C., Samain, D., Stramm, C., Noller, K., Müller, K. (2014). Water Repellence and Oxygen and Water Vapor Barrier of PVOH-Coated Substrates before and after Surface Esterification. *Polymers*, 6 (11), 2764–2783. doi:10.3390/polym6112764
- Buntinx, M., Willems, G., Knockaert, G., Adons, D., Yperman, J., Carleer, R., Peeters, R. (2014). Evaluation of the Thickness and Oxygen Transmission Rate before and after Thermoforming Mono- and Multi-layer Sheets into Trays with Variable Depth. *Polymers*, 6 (12), 3019–3043. doi:10.3390/polym6123019
- Kolosov, O. Ye. (2015). *Tekhnologiiia pakovalnoho vyrobnytstva [Technology of the packaging production]*. Kyiv: VPK «Politekhnik», 247.
- Kolosov, O. Ye. (2015). *Tekhnologiiia zberihannia kharchovoi produktsii: neupakovanoi ta upakovanoi iz zastosuvanniam polimernykh plickovykh materialiv [Storage Technology of food products: not packaged and packed with polymer film materials]*. Kyiv: VPK «Politekhnik», 179.
- Kolosov, O. Ye., Sidorov, D. E., Maletskiy, S. V. (2016). Proektuvannia pakovalnykh polimernykh plickovykh materialiv z neobkhdnykh bariernykh vlastyvostiamy. *Visnyk NTUU «KPI». Seriiia «Khimichna inzheneriia, ekolohiia ta resursozbezehennia*, 1 (15), 15–20.
- In: Harper, C. A. (2005). *Handbook of Plastic Processes*. John Wiley & Sons, Inc., 743. doi:10.1002/0471786586
- Massey, L. K. (2003). *Permeability Properties of Plastics and Elastomers*. NY: Plastics Design Library, 601. doi:10.1016/b978-1-884207-97-6.50103-0
- In: Knoll, W., Advincula, R. C. (2011). *Functional Polymer Films*. Wiley-VCH Verlag GmbH & Co. KGaA, 1080. doi:10.1002/9783527638482
- Wagner, J. R. Jr. (2010). *Multilayer Flexible Packaging*. Elsevier, 258. doi:10.1016/b978-0-8155-2021-4.10019-x
- Briston, J. H., Katan, L. L. (1988). *Plastics Films. Ed. 3*. Longman Scientific & Technical, 1050.
- Risch, S. (2000). Migration of Toxicants and Odoractive Substances from Flexible Packaging Materials to Food. *Journal of Food Technology*, 7, 95–102.
- Choudhry, M. S., Lox, F., Buekens, A., Decroly, P. (1998). Evaluation of migrational behaviour of plastic food-contact materials: a comparison of methods. *Packaging Technology and Science*, 11 (6), 275–283. doi:10.1002/(sici)1099-1522(199811/12)11:6<275::aid-pts441>3.0.co;2-r
- De Abreu, D. A. P., Cruz, J. M., Angulo, I., Losada, P. P. (2009). Mass transport studies of different additives in polyamide and exfoliated nanocomposite polyamide films for food industry. *Packaging Technology and Science*, 23 (2), 59–68. doi:10.1002/pts.879
- Baner, A. L., Piringer, O. (2007). Preservation of quality through packaging. *Plastic Packaging Materials for Food*. Wiley-VCH Verlag GmbH, 1–8. doi:10.1002/9783527613281.ch01

OBTAINING OF PURIFIED SOLUTIONS OF SODIUM HYDROXIDE FOR SYNTHESIS OF FERRATES(VI)

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One of the important problems in the production of ferrates(VI) is the presence of impurities of heavy metal compounds (HMC) in the feedstock, which reduce the yield and quality of target products. Therefore, for HMC removal from alkaline solution serving as a medium for the synthesis of Fe(VI), it is proposed to use pre-processing by a constant electric current.

The study of the processes occurring in the electrolysis of concentrated NaOH solution is the aim of this research.

During experiments it was found that most of impurity metals (Fe, Ni, Pb, Sn, Zn) are discharged at the cathode with the formation of metal deposits and therefore can be easily removed from the solution. It is shown that the speed and completeness of HMC removal are significantly different. Thus, the degree of chromium extraction does not exceed 5 %, and aluminum is not removed at all. Lead, iron and nickel are fastest removed from the system.

Research of the basic laws of electroextraction for a number of metals has allowed to establish the influence of the main factors (current density J , temperature T and electrolysis time τ) on the efficiency of removal of impurities. It was shown that with increasing current density increases the completeness of the removal of impurities. T increase leads to an increase in the recovery rate, which is obviously due to the decrease in electrolyte viscosity and increase of diffusion coefficient of the discharging particle.

Decrease in the rate of decomposition of the anion FeO_4^{2-} is recorded in sodium hydroxide treated with the proposed method. It confirms the feasibility of electroextraction in ferrate synthesis technologies.

Keywords: synthesis of ferrates(VI), sodium hydroxide, heavy metal impurities, purification, electrolysis.

References

- Sharma, V. K. (2008). *Ferrates: Synthesis, Properties, and Applications in Water and Wastewater Treatment*. ACS Symposium Series, 524. doi:10.1021/bk-2008-0985
- Yates, B. J., Zboril, R., Sharma, V. K. (2014). Engineering aspects of ferrate in water and wastewater treatment – a review. *Journal of Environmental Science and Health, Part A*, 49 (14), 1603–1614. doi:10.1080/10934529.2014.950924
- Gan, W., Sharma, V. K., Zhang, X., Yang, L., Yang, X. (2015). Investigation of disinfection byproducts formation in ferrate(VI) pre-oxidation of NOM and its model compounds followed by chlorination. *Journal of Hazardous Materials*, 292, 197–204. doi:10.1016/j.jhazmat.2015.02.037
- Jiang, J.-Q., Durai, H. B. P., Winzenbacher, R., Petri, M., Seitz, W. (2014). Drinking water treatment by in situ generated ferrate(VI). *Desalination and Water Treatment*, 55 (3), 731–739. doi:10.1080/19443994.2014.938303
- Veprek-Siska, J., Ettl, V. (1967). Reactions of very pure substances: Decomposition of Manganese(VII), Iron(VI) and Ruthenium(VII) oxyanions in alkaline solution. *Chemistry and Industry*, 1, 548–549.
- Licht, S., Wang, B., Ghosh, S. (1999). Energetic Iron(VI) Chemistry: The Super-Iron Battery. *Science*, 285 (5430), 1039–1042. doi:10.1126/science.285.5430.1039
- Golovko, D., Golovko, I. (2015). Effect of cobalt compounds on stability of ferrates(VI). *Technology Audit And Production Reserves*, 6(4(26)), 62–66. doi:10.15587/2312-8372.2015.56273
- Golovko, D. (2015). Effect of chromium compounds on synthesis of alkali metal ferrates (VI). *Eastern-European Journal Of Enterprise Technologies*, 3(6(75)), 15–21. doi:10.15587/1729-4061.2015.42634
- Joo, J., Kim, J., Kim, J. W., Ocon, J. D., Lee, J. K., Chang, W., Lee, J. (2015). Ultrahigh purification in concentrated NaOH by electrowinning for solar cell application. *Separation and Purification Technology*, 145, 24–28. doi:10.1016/j.seppur.2015.02.011
- Yuan, B., Kongstein, O. E., Haarberg, G. M. (2009). Electrowinning of Iron in Aqueous Alkaline Solution Using a Rotating Cathode. *Journal of The Electrochemical Society*, 156 (2), 64–69. doi:10.1149/1.3039998
- Chengchun, J., Chen, L., Shichao, W. (2008). Preparation of Potassium Ferrate by Wet Oxidation Method Using Waste Alkali: Purification and Reuse of Waste Alkali. *ACS Symposium Series*, 985, 94–101. doi:10.1021/bk-2008-0985.ch005
- Delaude, L., Laszlo, P. (1996). A Novel Oxidizing Reagent Based on Potassium Ferrate(VI). *The Journal of Organic Chemistry*, 61 (18), 6360–6370. doi:10.1021/jo960633p
- Golovko, D. A., Sharma, V. K., Suprunovich, V. I., Pavlova, O. V., Golovko, I. D., Bouzek, K., Zboril, R. (2011). A Simple Potentiometric Titration Method to Determine Concentration of Ferrate(VI) in Strong Alkaline Solutions. *Analytical Letters*, 44 (7), 1333–1340. doi:10.1080/00032719.2010.511748

PERMEABILITY EVALUATION OF NATURAL SAUSAGE CASINGS

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One of the main factors of inadequate use of natural casings is their high permeability, which leads to technological defects of finished products, substantial losses in the manufacturing and storage of sausages and economically making production less profitable. The problem of permeability adjusting of natural casings is related to its definition. Despite the changes in chemical composition and functional and technological properties of raw meat in recent years, significant differences in the gastrointestinal tract of animals, data on the permeability of the intestinal casings is currently available. In addition, methodology for permeability evaluation of natural sausage casings isn't developed, and known methods for artificial casings need to be improved in order to adapt them to the intestine.

Disadvantages of the known methods of determining water permeability, aroma permeability and fat permeability are identified as a result of the studies. The necessity is proved to develop measures to improve methods aimed to reproduce the conditions as realistic manufacturing technology and storage of sausages. The methods for determining aroma permeability and fat permeability of intestinal casings are improved and adapted. The proposed method of determining aroma permeability excludes a necessity for sealing of intestinal casings and ensures integrity of vessels, closed by intestine casings using an aluminum cap and circular PTFE membrane. The weighing method for determining fat permeability is improved because the pressure on pork fat deposited on a substrate pattern was created by placing the cargo, samples of intestinal casings were dried to constant weight before the tests. Fat permeability is measured at set of intervals.

The results of permeability evaluation of different kinds of natural sausage casings indicate their significant difference (in 2,1 times – for water and vapor permeability, in 1,6 times – for aroma permeability, in 1,9 times – for fat permeability), which leads to feasibility of differential approach to forecasting and rationing losses in storage technology of sausages in the intestinal casings. It is established that the main factors that cause the obtained data of intestinal casing permeability are kind of slaughter cattle from which they are derived, the degree of technological processing, anatomic origin (the components of the set of casings) and lifetime function.

Keywords: natural (intestinal) sausage casings, permeability, water permeability, aroma permeability, fat permeability, methods for determining permeability.

References

- Onishchenko, V. M., Shubina, L. Yu., Yancheva, M. O. (2009). *Naukovi ta praktychni aspekty vyrobnytstva i zastosuvannya naturalnykh kovbasnykh obolonok*. Kharkiv: KhDUKht, 149.
- Yancheva, M., Onishchenko, V., But, O. (2014, July). Prioritet – effektivnost'. Tendentsii rynku kolbasnykh obolonok. *Mir produktov*, 36–38.
- Aksenova, T. I., Ananiev, V. V., Dvoretkaia, N. M. et al.; In: Rozantsev, E. G. (2002). *Tehnologiya upakovochного proizvodstva*. Moscow: Kolos, 184.
- Gul', V. E., Beliatskaia, O. N. (1968). *Plenochnye polimernye materialy dlia upakovki pishchevnykh produktov*. Moscow: Pishchevaia promyshlennost', 278.
- Verghese, K., Lewis, H., Lockrey, S., Williams, H. (2015, April 9). Packaging's Role in Minimizing Food Loss and Waste Across the Supply Chain. *Packaging Technology and Science*, 28 (7), 603–620. doi:10.1002/pts.2127
- Ulitskii, Z. Z. (2002). Obolonchka «Poli-Pak» – nadezhnaia zashchita kolbasnykh izdelii. *Miasnaia industriia*, 2, 33–34.
- Levanichev, V. V. et al. (2003). Issledovanie bar'ernykh svoystv poliamidnykh obolonok. *Miasnoe delo*, 1, 10–11.

8. Patino, J. H., Henriquez, L. E., Restrepo, D., Mendoza, M. P., Lantero, M. I., Garcia, M. A. (2014). Evaluation of polyamide composite casings with silver-zinc crystals for sausages packaging. *Food Packaging and Shelf Life*, 1 (1), 3–9. doi:10.1016/j.fpsl.2013.09.001
9. Acerbi, F., Guillard, V., Guillaume, C., Gontard, N. (2016). Assessment of gas permeability of the whole packaging system mimicking industrial conditions. *Food Packaging and Shelf Life*, 8, 81–85. doi:10.1016/j.fpsl.2016.04.003
10. Schmid, M., Zillinger, W., Muller, K., Sangerlaub, S. (2015). Permeation of water vapour, nitrogen, oxygen and carbon dioxide through whey protein isolate based films and coatings – Permselectivity and activation energy. *Food Packaging and Shelf Life*, 6, 21–29. doi:10.1016/j.fpsl.2015.08.002
11. East, A. R., Samarakoon, H. C., Pranamornkith, T., Bronlund, J. E. (2015). A Review of Ethylene Permeability of Films. *Packaging Technology and Science*, 28 (8), 732–740. doi:10.1002/pts.2137
12. Felix, J. S., Manzoli, J. E., Padula, M., Monteiro, M. (2013). Evaluation of Different Conditions of Contact for Caprolactam Migration from Multilayer Polyamide Films into Food Simulants. *Packaging Technology and Science*, 27 (6), 457–466. doi:10.1002/pts.2046
13. Onishchenko, V. M., Yancheva, M. O., Ostroverkh, I. S. (2009). Khimichnyi sklad kyshok ta vmist u nykh toksychnykh elementiv. *Prohresyvni tekhnika ta tekhnolohii kharchovykh vyrobnytstvo restorannoho hospodarstva i torhivli*, 2 (10), 466–472.
14. Nagrodskii, Yu. R. (1967). *Upakovochnye materialy s pokrytiem iz mikrokristallicheskih voskov*. Moscow: Informstandartelektro, 39.
15. Onishchenko, V. M., Shubina, L. Yu., Yancheva, M. O., Ostroverkh, I. S., Chuiko, A. M., Shevchenko, V. H.; assignee: Kharkiv State University of Food Technology and Trade. (10.11.2010). Sposib vyznachennia aromatopronyknosti naturalnykh kovbasnykh obolonok. *Patent of Ukraine № 54388, MIIK A 22 C 17/00, A 22 C 13/00*. Appl. № u201004445. Filed 16.04.2010. Bull. № 21, 6.
16. Dergunova, A. A. (1976). *Obrabotka kishok*. Moscow: Pishchevaia promyshlennost', 174.
17. Onishchenko, V. M., Shubina, L. Yu., Ostroverkh, I. S. (2009). Analiz mekhanichnykh kharakterystyk naturalnykh kovbasnykh obolonok ta metodiv yikh vyznachennia. *Prohresyvni tekhnika ta tekhnolohii kharchovykh vyrobnytstvo restorannoho hospodarstva i torhivli*, 1 (9), 339–347.
18. Tatulov, Yu. V., Krehov, N. M., Sus', I. V. (2001). Ispol'zovanie konservantov pri hranenii natural'nykh kolbasnykh obolonok. *Hranenie i pererabotka sel'hozsyr'ia*, 10, 42–44.

INFLUENCE OF BOTTLE AGING ON THE DYNAMICS OF QUALITY INDICATORS OF WINES OF CONTROLLED NAMES OF ORIGIN

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The method of bottle aging is used in the production of still and sparkling wines of high quality, pre-aged in barriques and without it in various wine regions of the world. The result of the study is investigation of bottle aging effect on the dynamics of quality indicators of white table wines of controlled names of origin produced in LLC «Industrial-trading company Shabo» (Odessa region, Ukraine). The object of the study were prototypes of white table wine-making materials produced from grape of Chardonnay, Riesling and Telti Kuruk varieties according to technological instructions developed by LLC «ITC Shabo».

The main groups of aromatic components were defined, namely aliphatic aldehydes, dimethyl sulfides, acetaldehyde, aromatic aldehydes and aliphatic lactones, which were changed during the experiment. Prototypes of white table wines of controlled names of origin were prepared according to the technological instructions developed in the production of LLC «ITC Shabo» from Chardonnay, Riesling, Telti Kuruk varieties. Modes and bottle

aging parameters for white table wine were developed: 6 months at a temperature of 12–15 °C.

Analysis of the research results showed that the dominant flavor was furfural and 5-metilfurfural that may be indicative of more than 6 months potential of these wines in the bottle.

Keywords: bottle aging, aromatics, wine of controlled names of origin, quality.

References

1. Pineau, N., Schlich, P., Cordelle, S., Mathonniere, C., Issanchou, S., Imbert, A., Rogeaux, M., Etievant, P., Koster, E. (2009). Temporal Dominance of Sensations: Construction of the TDS curves and comparison with time-intensity. *Food Quality and Preference*, 20 (6), 450–455. doi:10.1016/j.foodqual.2009.04.005
2. Meillon, S., Viala, D., Medel, M., Urbano, C., Guillot, G., Schlich, P. (2010). Impact of partial alcohol reduction in Syrah wine on perceived complexity and temporality of sensations and link with preference. *Food Quality and Preference*, 21 (7), 732–740. doi:10.1016/j.foodqual.2010.06.005
3. Guinard, J.-X., Pangborn, R. M., Lewis, M. J. (1986). The Time-Course of Astringency in Wine upon Repeated Ingestion. *American Journal of Enology and Viticulture*, 37, 184–189
4. Carey, V. A., Archer, E., Saayman, D. (2003). Landscape diversity in Stellenbosch: Implications for viticulture. In: *Proc. Colloquium International Paysage de Vignes et du Vins: Patrimoine – Enjeux – Valorisation, July 2003, Fontevraud, France*. Fontevraud, 112–117.
5. *Proceedings of the 36th World Congress of Vine and Wine «Vine and Wine Between Tradition and Modernity»*. (2013). Bucharest, Romania, 252. ISBN 979-10-91799-15-7.
6. Ventura, F., Rossi Pisa, P., Vicari, A. (2004). Effect of land use on soil erosion in a small watershed of Emilia Romagna region. *Italian Journal of Agronomy*, 8 (1), 29–36.
7. *Polozhennia pro vynohradni vyna kontrolovanykh naimenuvan za pokhodzhenniam (KNP) KD U 37471967-11.02-3:2012*. (2012). Ministry of Agrarian Policy and Food of Ukraine, 12.
8. *Metodyka kontroliu yakosti vynohradu, protsesu vyrobnytstva, yakosti ta identyfikatsii vynohradnykh vyn kontrolovanykh naimenuvan za pokhodzhenniam (KNP) KD U 37471967-11.02-4:2012*. (2012). Ministry of Agrarian Policy and Food of Ukraine, 14.
9. In: Gerzhikova, V. G. (2009). *Metody tehnohimicheskogo kontroliu v vinodelii*. Ed. 2. Simferopol: Tavrida, 304.
10. Kishkovskii, Z. N., Merzhanian, A. A. (1984). *Tehnologiya vina*. Moscow: Legkaia i pishchevaia promyshlennost', 504.
11. Reitblat, B. B. (2002). Holodnyi steril'nyi rozliv dlia povysheniia kachestva i stabil'nosti vin. *Vinodelie i vinogradarstvo*, 5, 22–23.
12. Shaub, H. P. (2001). Holodnyi rozliv tihiv vin. *Vinodelie i vinogradarstvo*, 4, 20–21.
13. Ribereau-Gayon, P., Dubourdieu, D., Doneche, D. (2012). *Traite d'oenologie. T. 2: Chimie du vin – Stabilisation et traitements*. Ed. 6. Dunod, 624.
14. Usseglio-Tomasset, L. (1995). *Chimie Oenologique*. Ed. 2. Paris: Technique & Documentation, 332.
15. Pisarnitskii, A. F. (1994). Atsetal'degid – odin iz faktorov okisleniia etanola. *Vinograd i vino Rossii*, 1, 27–28.
16. Pisarnitskii, A. F. (2002). AO-geterotsiklicheskie soedineniia v aromate vinodel'cheskoi produktsii. *Vinogradarstvo i vinodelie*, 3, 22–23.
17. Popov, K. S., Chistiakova, N. P. (1962). *Tehnologiya belyh stolovykh vin, obespechivaiushchaia povyshennuiu stabil'nost' protiv pomutnenii i izlishnei okislennosti*. Moscow: Sel'hozizdat, 252.
18. Dubourdieu, D.; In: Guimberto, G. (1992). Vinification des vins blancs secs en barriques. *Le bois et le qualite des vins et des eaux-de-vie*. Bordeaux: Vigne et vin publications internationales, 137–143.
19. Gunata, Y. Z., Sapis, J., Moutounet, M. (1987). Substrates and aromatic carboxylic acid inhibitors of grape phenol oxidases. *Phytochemistry*, 26 (6), 1573–1575. doi:10.1016/s0031-9422(00)82247-5

RESEARCH OF EFFECT OF CALCIUM URONIC COMPLEXES ON THE BASIC QUALITY INDICATORS OF THE CURD

page 32–36

The content of free calcium in order to implement the chemical potentials of ionotropic gelling agent and Ca²⁺ ions, which are the part of the milk-protein complex in the system of the liquid phase of cottage cheese, is investigated. Water-holding capacity (WHC) of food systems based on cottage cheese and calcium alginate gel is investigated to determine the concentration of a rational ionotropic polysaccharide solution. Calcium alginate gel is formed between anions of sodium alginate uronic acids and calcium ions contained in the liquid fraction of cottage cheese. Conservation modes of the food systems based on cottage cheese and uronic complexes are determined. It is proved that the use of sodium alginate ionotropic polysaccharide in cottage cheese-based semi-finished products will create a completely new product. This product will be based on the principle of controlled ionotropic gelation, which is an innovative technological solution to the problem of water holding in CCPC system. It is important to extend the shelf life and regeneration of technological functions after conservation while maintaining the high quality performance of cottage cheese and culinary products based on it. And there are no changes in the physical state, structure, organoleptic properties and nutritional value.

Keywords: cottage cheese, ionotropic gelation, quality, uronates, calcium, water-holding capacity.

References

1. Proteins, peptides and amino acids. (03.05.2011). *AzaQuar.com. Food Science and Technology*. Available: <http://www.azaquar.com/en/doc/proteins-peptides-and-amino-acids/>
2. Franks, F. (1985). Water and Aqueous Solutions: Recent Advances. *Properties of Water in Foods*. Springer Science + Business Media, 1–23. doi:10.1007/978-94-009-5103-7_1
3. Water in food. (03.05.2011). *AzaQuar.com. Food Science and Technology*. Available: <http://www.azaquar.com/en/doc/water-in-food/>
4. Bannikova, L. A., Koroliovna, N. S., Semeniuhina, V. F. (1987). *Mikrobiologicheskie osnovy molochnogo proizvodstva: Handbook*. Moscow: Agropromizdat, 400.
5. In: Fox, P. F., McSweeney, P. L. H., Cogan, T. M., Guinee, T. P. (2004). *Cheese: Chemistry, Physics and Microbiology. Vol. 1. General Aspects*. Elsevier Ltd., 617. doi:10.1016/s1874-558x(13)70001-0
6. Fox, P. F., Guinee, T. P., Cogan, T. M., McSweeney, P. L. H. (2017). *Fundamentals of Cheese Science*. New York: Springer, 730. doi:10.1007/978-1-4899-7681-9
7. Filchakova, N. N. (2002). Biotekhnologii i bezopasnost' pishchevykh produktov. *Pererabotka moloka*, 11, 4–5.
8. Tsintsadze, T. D., Lee, C.-H., Rha, C. (1978). Microstructure and Mechanical Properties of Single Cell Protein Curd. *Journal of Food Science*, 43 (2), 625–635. doi:10.1111/j.1365-2621.1978.tb02370.x
9. Davidson, P. M., Schmidt, S. E., Taylor, T. M. (2013). Chapter 30: Chemical Preservatives and Natural Antimicrobial Compounds. *Food Microbiology*. American Society for Microbiology, 765–801. doi:10.1128/9781555818463.ch30
10. Sarafanova, L. A., Kostrova, I. B. (1997). Uvelichenie srokov hraneniia molochnoi produktsii. *Hranenie i pererabotka sel'skohoziastvennogo syria*, 11, 39–41.
11. Zobkova, Z. S. (2005). Tehnologicheskie i tehnichekieskie resheniia povysheniia stoikosti v hranenii bioaktivnykh molochnykh produktov. *Molochnaia promyshlennost'*, 3, 38–43.
12. Kampf, N. (2000). Hydrocolloid coating of cheeses. *Food Hydrocolloids*, 14 (6), 531–537. doi:10.1016/s0268-005x(00)00033-3
13. Nechaev, A. P., Kochetkova, A. A., Zaitsev, A. N. (2002). *Pishchevye dobavki: Handbook*. Moscow: Kolos-Press, 256.
14. Olsen, S. (2002). Rol' stabilizatorov v proizvodstve kislomolochnykh produktov. *Molochnaia promyshlennost'*, 8, 32–33.

15. Belov, V. V., Noskov, A. V. (1994). Proizvodstvo tvorozhnykh izdelii i iogurtov s ispol'zovaniem stabilizatsionnykh sistem. *Molochnaia promyshlennost'*, 2, 26–27.
16. Zobkova, Z. S., Fursova, T. P. (2003). O konsistentsii kislomolochnykh produktov. *Molochnaia promyshlennost'*, 1, 49–51.
17. Pyvovarov, Y. P. (2009). Kinetyka protsesu strukturuvannia kharchovykh modelnykh system na osnovi ionotropnoho polisakharydu natriu alhinatu. *Tovary i rynky*, 2, 48–58.

RESEARCH OF TECHNOLOGICAL PROPERTIES OF GRAPE SKIN POWDER AS AN FUNCTIONAL INGREDIENT OF MAYONNAISE SAUCE

page 36–41

The improvement of the technology of emulsion foods as the most consumed by all groups of the Ukrainian population, by introducing as a functional ingredient, as well as capable of influencing the rheological characteristics of vegetable raw material powder was proposed. The object of this research is grape skin powder, which is a source of natural antioxidants polyphenol nature, dietary fiber and pectin substances. One of the basic requirements applicable to mayonnaise production is to obtain a stable emulsion having optimum viscosity. This is achieved by introducing into the formulation the starches as thickening agents. However, the production of mayonnaise products of health improvement requires a presence in formulation components having a high biological and physiological value. It is proposed to add into the mayonnaise sauce grape skin powder as an ingredient with specific technological properties. This technological decision will allow to change the rheological properties of mayonnaise products, and will give it the wellness status. The technological properties of grape skin Black Pearl powder, namely the water-holding capacity, the effective viscosity and surface tension were studied. Determination of these properties of grape skin powder was conducted on the respective equipment and by standard techniques. The influence of pre-treatment temperature and acid concentration on changes in terms of technological properties of the samples suspensions of grape skin powder was analysed. The optimal parameters (the temperature and acid concentration) to convert of protopectin in the soluble state and an increase in the water retention capacity, the effective viscosity and surface tension of sample suspensions of grape skin Black Pearl powder was proposed. It is proved technological solution proposed by introduction of a plant raw material powder in the formulation of mayonnaise sauces for the possibility of changing the rheological properties and assigning these products to health improvement food status.

Keywords: grape skin powder, technological properties, mayonnaise sauce, acid concentration, temperature.

References

1. Smoliar, V. I. (2007). Osnovni tendentsii v kharchuvanni nase-lennia Ukrainy. *Problemy kharchuvannia*, 4. Available: http://www.medved.kiev.ua/arh_nutr/art_2007/n07_4_1.htm
2. Nishinari, K. (2008). Structure and Properties of Food Hydrocolloids – Gels, Emulsions and Foams. *Foods Food Ingredients J. Jpn.*, 213 (5), 138–141.
3. Milani, J., Maleki, G. (2012). Hydrocolloids in Food Industry. *Food Industrial Processes – Methods and Equipment*, 2, 17–38. doi:10.5772/32358
4. Yokohira, M., Yamakawa, K., Saoo, K., Matsuda, Y., Hosokawa, K., Hashimoto, N., Kuno, T., Imaida, K. (2008). Antioxidant Effects of Flavonoids Used as Food Additives (Purple Corn Color, Enzymatically Modified Isoquercitrin, and Isoquercitrin) on Liver Carcinogenesis in a Rat Medium-Term Bioassay. *Journal of Food Science*, 73 (7), 561–568. doi:10.1111/j.1750-3841.2008.00862.x

5. Frey, J. A. (2005). The Effects of Sugar Acidity and Pectin on Gel Strength in a Naturally Low and High Pectin Fruit Varieties. *Food Chemistry*. Purdue University, 24. Available: http://www.cfs.purdue.edu/fn/fn453/Project_Archive/Fall_2005/Variables_affecting_get_strength_in_pectin_jellies.pdf
6. Thomas, R. L. (2007). Hydrocolloids: Fifteen Practical Tips. *Guaranteed Gums*, 8, 2–17.
7. Shchedushnov, D. E., Perov, V. N., Melnik, N. V. (2002). Stabil'naia emul'siia – otlichnyi maionez. *Masla i zhiry*, 12 (22), 18–19.
8. DSTU 4560:2006. *Maionezy. Pravyly pryjimannia ta metody vyprovuvannia*. (2008). Introduced: 2008-01-01. Kyiv: Derzhspozhyvstandart Ukrainy, 16.
9. Volkov, V. A. (2015). *Kolloidnaia himiia. Poverhnostnye iavleniia i dispersnye sistemy*. Ed. 2. Saint Petersburg: Lan, 672.
10. Davydenko, N. V., Smyrnova, I. P., Horbas, I. M., Kvasha, O. O. (2002). Neratsionalne kharchuvannia – ryzyk dlia zdorovia. *Ukrainskyi terapevtychnyi zhurnal*, 3, 26–29.
11. World Health Organization Regional Office for Europe. (2014 August 27). Progress Report on the European Environment and Health Process: EUR/RC6 4/24 Rev.1. *Sixty-Fourth Session of the Regional Committee for Europe, Copenhagen, Denmark, September 15-18, 2014*. Available: <https://www.unece.org/fileadmin/DAM/env/cep/CEP-20/eur.rc64.24.rev.1.r.pdf>
12. In: Vlasenko, N. S. (2014). *Balansy ta spozhyvannia osnovnykh produktiv kharchuvannia naseleenniia Ukrainy. Statystychnyi zbirnyk 2013*. Kyiv: State Statistics Service of Ukraine, 59.
13. Ananieva, V. V., Krychkovska, L. V., Belinska, A. P., Petrov, S. A. (2016). Improvement of antioxidant stability of oil base of emulsion products for health improvement purposes. *Bulletin of NTU «KhPI»*. Series: *Innovation researches in students' scientific work*, 19 (1191), 75–81.
14. Dunets, E. G., Zaiko, G. M., Bedilo, M. S. (2008). Vliianie tehnologicheskikh faktorov na reologicheskie svoistva sousov funktsional'nogo naznacheniia. *Izvestiia vuzov. Pishchevaia tehnologiya*, 4, 50–52.
15. Elisheeva, N. E. (2008). *Razrabotka tehnologii funktsional'nykh zhirovykh produktov emul'sionnoi prirody s pishchevymi voloknami i biologicheski aktivnymi veshchestvami*. Moscow, 176.
16. Solopova, A. N. (2006). *Razrabotka i issledovanie tehnologii maionezov s produktami pererabotki tykvy*. Kemerovo, 159.
17. Smychagin, O. V. (2009). *Razrabotka retseptur i issledovanie kachestva dieticheskikh maioneznych sousov s primeneniem produktov pererabotki zarodysheh kukuryzy*. Krasnodar, 164.
18. Liu, H., Xu, X. M., Guo, S. D. (2007). Rheological, texture and sensory properties of low-fat mayonnaise with different fat mimetics. *LWT – Food Science and Technology*, 40 (6), 946–954. doi:10.1016/j.lwt.2006.11.007
19. Lobanov, V. G., Shcherbakov, V. G., Prudnikova, T. N. et al. (2010). *Laboratornyi praktikum po biohimii i pishchevoi himii*. Ed. 2. Krasnodar: Kuban State University of Technology, 213.
20. Kovalevska, Ye. I., Serbova, M. I., Volovyk, L. S., Tymokhin, V. V. (2001). *Metodychni vkazivky do vyvchennia rozdilu «Strukturno-mekhanichni vlastyvoosti dyspersnykh sistem»*. Kyiv: UDUKht, 282.

COMMODITY ANALYSIS OF SPREADS WITH INCREASED BIOLOGICAL VALUE

page 42–46

Two samples of spreads from different manufacturers that are sold in the trade network of Kharkov (Ukraine) are selected as the object of research – «Poltavchanka» and «Farmer», and water-oil extract of oats seedlings. The samples were manufactured in accordance with DSTU 4445:2005.

For obtaining a new spread specie with high biological activity and stable emulsion in each sample at room temperature was introduced certain amount of oats seedling extract (10 %, 20 %, 30 %, 40 %, 50 %) depending on the amount of sample weight, respectively. The disadvantage of this method is the use of

energy-intensive equipment, high labor input in the preparation of raw materials.

Research of the properties of raw materials, model samples of spreads and finished products was carried out by conventional and modern methods, including chromatography, spectrophotometry, and cytomorphologobiophysical diagnosis. Mathematical and statistical processing of the experimental results was performed using PC in environment MS Excel 2007.

Research proved the feasibility of use and rational concentration of natural antioxidant capable with immunostimulatory action, water-oil extract of oats seedlings in obtained new spread for the purpose of inhibition of oxidative and hydrolytic processes and acceleration of spread digestion time and metabolism acceleration. The complexity in the preparation of raw materials was also reduced by reducing manufacturing operations.

Thanks to the addition of this extract it is expected to get the biologically optimal formula in balance of fatty acids and polyunsaturated fatty acids of ω_3 and ω_6 groups with high content of essential α -linolenic acid of ω_3 group, carotenoids, tocopherols and low content of trans-isomers of fatty acids.

Keywords: spreads, water-oil extract of oats seedlings, commodity science, increase of biological and nutritional value.

References

1. Syrokhman, I. V., Rodak, O. Ya. (2005). Tovaroznavcha otsinka sprediv riznykh hrup. *Naukovyi visnyk Lvivskoho natsionalnoho universytetu veterinarnoi medytsyny ta biotekhnologii imeni S. Z. Gzhytskoho*, 7(4(27)), Part 1, 134–138.
2. Syrokhman, I. V., Rodak, O. Ya. (2008). Sprediv funktsionalnoho pryznachenniia. *Naukovyi visnyk Lvivskoho natsionalnoho universytetu veterinarnoi medytsyny ta biotekhnologii imeni S. Z. Gzhytskoho*, 10(2(37)), Part 3, 167–170.
3. Rodak, O. Ya. (2008). Sprediv pidvyshchenoi biolohichnoi tsinosti. *Naukovyi visnyk Lvivskoho natsionalnoho universytetu veterinarnoi medytsyny ta biotekhnologii imeni S. Z. Gzhytskoho*, 10(3(38)), Part 3, 367–370.
4. Semakin, F. N., Belenko, E. L.; assignee: Moscow State University of Food Production. (10.06.2010). Rastitel'no-zhirovoi spred. *Patent of Russia № 2391019*. Appl. № 2009121759/13. Filed 09.06.2009.
5. Kalmanovich, S. A., Drozdov, A. N., Brikota, T. B.; assignee: Kuban State University of Technology. (10.12.2009). Spred dieticheskii. *Patent of Russia № 2374853*. Appl. № 2008118627/13. Filed 12.05.2008. Available: <http://partkom.com/patent/ru2374853/>
6. Hrek, O. V., Savchenko, O. A., Tymchuk, A. V., Diatel, O. V.; assignee: National University of Food Technologies. (25.01.2011). Composition of spread with medicine plant products. *Patent of Ukraine № 56645*. Appl. № u201007201. Filed 10.06.2010. Bull. № 2. Available: <http://uapatents.com/3-56645-sklad-spredu-z-produktami-likarsko-roslini.html>
7. Hrek, O. V., Savchenko, O. A., Onopriichuk, O. O., Petryna, A. B., Tymchuk, A. V.; assignee: National University of Food Technologies. (10.07.2009). Method for production of spread with filler. *Patent of Ukraine № 87382*. Appl. № u200712801. Filed 19.11.2007. Bull. № 13. Available: <http://uapatents.com/3-87382-sposib-virobnictva-spredu-z-napovnyuvachem.html>
8. Loseva, A. I. (2006). *Razrabotka i issledovanie tehnologii shivochno-rastitel'nogo sprediva s antioksidantnymi svoistvami*. Kemerovo, 20.
9. Do, V. H., Mun, S., Kim, Y.-L., Rho, S.-J., Park, K. H., Kim, Y.-R. (2016). Novel formulation of low-fat spread using rice starch modified by 4- α -glucanotransferase. *Food Chemistry*, 208, 132–141. doi:10.1016/j.foodchem.2016.03.101
10. Shen, Y., Kennedy, O. B., Methven, L. (2017). The effect of genotypical and phenotypical variation in taste sensitivity on liking of ice cream and dietary fat intake. *Food Quality and Preference*, 55, 79–90. doi:10.1016/j.foodqual.2016.08.010
11. Kar, S., Bhattacharyya, D. K., Ghosh, M. (2016). Evaluation of Nutritional Characteristics of Health Beneficial DAG Rich oil based Spreads. *Materials Today: Proceedings*, 3 (10), 3375–3387. doi:10.1016/j.matpr.2016.10.019

12. Rodak, O. Ya. (2009). Polipshennia zhyrnokyslotnoho skladu sprediv z vykorystanniam netradytsiinykh olii. *Naukovi pratsi ONAKhT*, 2 (36), 149–152.
13. Rodak, O. Ya. (2008). Vplyv emulhatoriv na yakist i stabilnist sprediv. *Prohresyeni tekhnika ta tekhnologii kharchovykh vyrobnytstvo restorannoho hospodarstva i torhivli*, 1 (7), 365–368.
14. Rodak, O. Ya. (2009). Vykorystannia ekstraktiv na osnovi likarsko-tekhnichnoi syrovyny yak naturalnykh antyoksydantiv dlia sprediv. *Tezy dopovidei III mizhhaluzevoi mizhnar. nauk.-prakt. konf. «Kharchovi dobaVKy. Kharchuvannia zdorovoi ta khvoroi liudyny»*, Donetsk, 12-13 bereznia 2009. Donetsk: DonNUET, 116–118.
15. Shahbazov, S. V. (2009). Zavisimost' elektrootritsatel'nosti kletochnykh iader cheloveka ot rezhima elektroforeza i vozrasta donorov. *Visnyk problem biologii i medytsyny*, 2, 25–30.

3. Horobets, S. V., Nhuien, T. Z., Karpenko, Yu. V. (2012). Doslidzhennia sorbtsii ioniv zaliza mahnitomichenym biosorbentom. *VI Vseukrainska naukovo-praktychna konferentsiia «Biotekhnologii XXI stolittia»*, 5 kvitnia 2012. Kyiv: NTUU «KPI», 147.
4. Patzak, M., Dostalek, P., Fogarty, R. V., Safarik, I., Tobin, J. M. (1997). Development of magnetic biosorbents for metal uptake. *Biotechnology Techniques*, 11 (7), 483–487. doi:10.1023/a:1018453814472
5. Horobets, S. V., Horobets, O. Yu., Dvojenko, O. K. et al. (2010). Ochyshchennia stichnykh vod vid ioniv Cu²⁺ (II) mahnitokerovanyim biosorbentom za dopomohoiu vysokohradiientnykh feromahnitnykh nasadok. *Naukovi visti NTUU «KPI»*, 3, 21–25.
6. Horobets, S. V., Karpenko, Yu. V., Marynchenko, L. V. (2010). Biosorbtsiia ioniv midi Cu²⁺ mahnitomichenymy klitynamy *S.serevisiae*. *Visnyk Donetskoho natsionalnoho universytetu. Ser. A. Pryrodnychi nauky*, 1, 230–236.
7. Yang, S. H., Lee, T., Seo, E., Ko, E. H., Choi, I. S., Kim, B.-S. (2011). Interfacing Living Yeast Cells with Graphene Oxide Nanosheets. *Macromolecular Bioscience*, 12 (1), 61–66. doi:10.1002/mabi.201100268
8. Peng, Q., Liu, Y., Zeng, G., Xu, W., Yang, C., Zhang, J. (2010). Biosorption of copper(II) by immobilizing Saccharomyces cerevisiae on the surface of chitosan-coated magnetic nanoparticles from aqueous solution. *Journal of Hazardous Materials*, 177 (1–3), 676–682. doi:10.1016/j.jhazmat.2009.12.084
9. Horobets, S. V., Karpenko, Yu. V., Kovalov, O. V., Olishchuk, V. V. (2013). Zastosuvannia mahnitomichenykh klityn *S.cerevisiae* yak biosorbentu na ochysnykh sporudakh. *Naukovi visti NTUU «KPI»*, 3, 42–47.
10. Safarik, I., Maderova, Z., Pospiskova, K., Horska, K., Safarikova, M. (2014). CHAPTER 10. Magnetic Decoration and Labeling of Prokaryotic and Eukaryotic Cells. *RSC Smart Materials*. London: Royal Society of Chemistry (RSC), 185–215. doi:10.1039/9781782628477-00185
11. Safarik, I., Maderova, Z., Pospiskova, K., Baldikova, E., Horska, K., Safarikova, M. (2015). Magnetically responsive yeast cells: Methods of preparation and applications. *Yeast*, 32 (1), 227–237. doi:10.1002/yea.3043
12. Wu, H. Q., Wu, Q. P. (2013). Research Progress of Nanomaterials about Removal of Toxic Metal Ions and Organics Used in Water Treatment. *Advanced Materials Research*, 662, 207–213. doi:10.4028/www.scientific.net/amr.662.207
13. Safarik, I., Pospiskova, K., Baldikova, E., Safarikova, M. (2016). Magnetically Responsive Biological Materials And Their Applications. *Advanced Materials Letters*, 7 (4), 254–261. doi:10.5185/amlett.2016.6176
14. Jadidian, R., Parham, H., Haghtalab, S., Asrarian, R. (2013). Removal of Copper from Industrial Water and Wastewater Using Magnetic Iron Oxide Nanoparticles Modified with Benzotriazole. *Advanced Materials Research*, 829, 742–746. doi:10.4028/www.scientific.net/amr.829.742
15. Infante, J. C. (2014). Removal of lead, mercury and nickel using the yeast *Saccharomyces cerevisiae*. *Revista MVZ Córdoba*, 19 (2), 4141–4149. Available: <http://revistas.unicordoba.edu.co/revistas/index.php/revistamvz/article/view/107>
16. Nguyen, M. L., Juang, R.-S. (2015). Improved biosorption of phenol using crosslinked chitosan beads after modification with histidine and *Saccharomyces cerevisiae*. *Biotechnology and Bioprocess Engineering*, 20 (3), 614–621. doi:10.1007/s12257-015-0039-7
17. Horobets, S. V., Chyzh, Yu. M., Kovalov, O. V., Shpetnyi, I. O. (2015). Efficiency of Magnetically Labelled Biosorbent Based on *Saccharomyces cerevisiae* Yeast for Sewage Treatment. *Naukovi visti NTUU «KPI»*, 3, 14–22
18. Horobets, S. V., Horobets, O. Yu., Chyzh, Yu. M., Kovalov, O. V.; assignee: National Technical University of Ukraine «Kyiv Polytechnic Institute». (25.08.2015). Sposib otrymannia mahnitokerovanoho biosorbentu. *Patent of Ukraine № 101016*. Appl. № u 2015 00909. Filed 05.02.2015. Bull. № 16. Available: <http://uapatents.com/5-101016-sposib-otrimannya-magnitokerovanogo-biosorbentu.html>

PRACTICAL USE OF DRY MAGNETICALLY CONTROLLED BIOSORBENT IN PURIFICATION SYSTEM OF DOMESTIC SEWAGE

page 46–51

The object of research is domestic sewage of sewage treatment facilities in Slavutich (Ukraine) and sorption capacity of the new dry magnetically controlled biosorbent (MCB) in relation to conventional pollutants of domestic sewage. The research found that the existing problem in wastewater treatment – exceeding MPD standards for indicators such as COD, ammonia nitrogen, nitrite, total iron, phosphates, smell – can be solved by using dry modified biosorbent obtained by magnetohydrodynamic mixing (MHDM) in crossed electric and magnetic fields. It was also selected optimal dose of dry MCB, which is 4 g/dm³ and optimal biosorption time – 20 minutes. The optimum particle size of dry MCB for effective biosorption is 0,1 mm. Efficiency of removal was determined not only for heavy metal ions by dry MCB on the basis of *Saccharomyces CEREVISIAE* yeast, obtained by MHDM in crossed electric and magnetic fields, but also for effective removal of such indicators of domestic sewage as: smell, COD, ammonia nitrogen, nitrite, phosphate, total iron.

As a result of researches it was found that dry MCB completely neutralizes the smell of sewage from the V points to 0 points. The effect of COD treatment is 48,1%; ammonia nitrogen – 46,7%; nitrite – 91,7%; phosphates – 64,4%; total iron – 51,7%.

Efficiency of dry MCB is in reduction of concentration of pollutants to MPD level. The advantage of dry MCB, obtained by MHDM in crossed electric and magnetic fields, is in storage and transportation. Research is useful because dry MCB, obtained by MHDM in crossed electric and magnetic fields, can be removed in speed mode using magnetic separation.

The results of this research can be used for dose selection of dry biosorbent for wastewater sewage of sewage treatment facilities, where there is a problem with the purification of the above parameters. This will allow treatment facilities to purify sewage water to MPD standards and reduce the negative impact on the reservoir.

Keywords: dry magnetically controlled biosorbent, biosorption, magnetite nanoparticles, wastewater, magnetohydrodynamic mixing.

References

1. Kovalov, O. V., Ivanova, I. M. (2010). Laboratorni doslidzhennia ozonuvannia stichnykh vod. *Zbirnyk naukovykh prats za materialamy VI mizhnarodnoi naukovo-praktychnoi konferentsii*. Chernihiv: ChDEIU, 158–161.
2. Wang, J., Chen, C. (2009). Biosorbents for heavy metals removal and their future. *Biotechnology Advances*, 27 (2), 195–226. doi:10.1016/j.biotechadv.2008.11.002

19. KND 211.1.4.034-95. *Metodyka fotometrychnoho vyznachennia zahalnoho zaliza z ortofenantrolinom v poverkhnevyykh i stichnykh vodakh.* (1995). Kyiv: Ministry of Environmental Protection of Ukraine, 10.
20. KND 211.1.4.030-95. *Metodyka fotometrychnoho vyznachennia amonii-ioniv z reaktyvom Neslera v stichnykh vodakh.* (1995). Kyiv: Ministry of Environmental Protection of Ukraine, 10.
21. KND 211.1.4.023-95. *Metodyka fotometrychnoho vyznachennia nitryt-ioniv z reaktyvom Hrissa v poverkhnevyykh ta ochyshchenykh stichnykh vodakh.* (1995). Kyiv: Ministry of Environmental Protection of Ukraine, 10.
22. KND 211.1.4.021-95. *Metodyka vyznachennia khimichnoho spozhyvannia kysniu (KhSK) v poverkhnevyykh i stichnykh vodakh.* (1995). Kyiv: Ministry of Environmental Protection of Ukraine, 10.
23. MVV 081/12-0005-01. *Poverkhnevi ta ochyshcheni stichni vody. Metodyka vykonannia vymiruvan masovoi kontsentratsii rozchynenykh ortofosfativ fotometrychnym metodom (0,05–100 mg/dm³).*
24. Lurie, Yu. Yu. (1984). *Analiticheskaia himiia promyshlennykh stochnykh vod.* Moscow: Himiia, 448.
25. Ivanova, Yu., Zenkin, A., Fedorchenko, Yu., Maziuk, N. (2012). Security assessment of major phases of the life cycle of wastewater regulation document. *Eastern-European Journal Of Enterprise Technologies*, 3(6(57)), 56–61. Available: <http://journals.urau.ua/ejet/article/view/4044>

DETERMINATION OF FLAME IONIZATION GAS ANALYZER ERROR

page 52–56

Flame ionization method is developed for automatic control of hydrocarbons in the atmosphere of the city in real time. This method measures the amount of concentration of hydrocarbon vapors and gases.

In modern gas analyzers of hydrocarbons, excluding methane, there are following main basic schemes:

1. Differential measurement scheme with two detectors and one amplifier. One detector gives a signal to the amount of hydrocarbons, other – only to methane.

2. Measurement scheme with one detector and one amplifier. Pneumatic scheme works by switching the flow of analyzed air into two streams, one of which is the flow of air through the device of methane separation in each measurement cycle.

3. The scheme contains two detectors, two gas blocks, one of which is equipped with a separation column of hydrocarbons and two measuring blocks.

Depending on the conditions and purpose of the application, it is necessary to select the right measuring system of flame ionization analyzer. Determination of the main sources of errors in measuring circuits of flame ionization gas analyzers with various basic schemes allowed conducting theoretical calculation of total errors, roughly estimating their size and selecting the optimal measuring circuit.

According to theoretical calculations, the basic error of differential measurement scheme with two electrometric amplifiers is 7,7 %. The values of basic errors of differential measurement scheme with one amplifier and scheme with one detector and adsorbing device are close to each other and are 10,26 % and 10,24 % respectively. The last measurement scheme, however, is characterized by large additional error due to the influence of often useless measurement parameters on the signal of flame ionization detector and is 18,63 %.

It is determined that the most optimal from considered gas analyzer schemes with the least basic error is differential measurement scheme with two electrometric amplifiers.

Keywords: measurement of hydrocarbons, methane, detector, flame ionization analyzer, errors of measurement channels of gas analyzers.

References

1. Tangiev, B. B. (2010). *Nauchnyi ekologo-kriminologicheskii kompleks (NEKK) po obespecheniu ekologicheskoi bezopasnosti i protivodeistviu ekoprestupnosti.* Saint Petersburg: Yuridicheskii tsentr – Press, 515.
2. Prymiskyi, V. P., Zhuzha, A. V. (2013). Stan ta perspektyvy rozvytku polumiano-ionizatsiinoho rozvytku dlia vymiruvannia kontsentratsii vuhlevodniv. *Metrolohiia ta prylady*, 2, 45–52.
3. Ryzhkov, V. F. (2004). Portativnyi plamenno-ionizatsionnyi gazoanalizator. *Tehnologia i konstruirovanie v elektronnoi apparature*, 1, 27–29.
4. Prymiskyi, V. P., Rumbeshta, V. A.; assignee: Private Joint Stock Company «Ukrainian Scientific Research Institute of Analytical Instrumentation». (16.09.2002). Flame-ionization analyzer for ecological monitoring of motor vehicles. *Patent of Ukraine № 49063 G01N 2762*. Appl. № 99116303. Filed 19.11.1999. Bull. № 9. Available: <http://uapatents.com/3-49063-polumyano-ionizatsijnij-gazoanalizator-ekologichnogo-kontrolyu-avtomobiliv.html>
5. Zhuzha, A. V. (10.07.2014). Flame-ionization analyzer for ecological monitoring of motor vehicles. *Patent of Ukraine № 91533 G01N 2762*. Appl. № u 2014 00559. Filed 20.01.2014. Bull. № 13. Available: <http://uapatents.com/4-91533-shvidkodyuchij-polumyano-ionizatsijnij-gazoanalizator.html>
6. Ivasenko, V. M., Prymiskyi, V. P., Korniienko, D. H., Vata-vu, A. V., Zhuzha, A. V. (2012). Suchasni zasoby instrumentalnoho kontroliu (hazoanalizatory i hazoanalitichni systemy) vidpratsovanykh haziv avtomobiliv. *Visnyk Natsionalnoho tekhnichnoho universytetu «KhPI»*. Ser. *Novi rishennia v suchasnykh tekhnolohiiakh*, 68, 135–141.
7. Bombaugh, K. J., Freeman, W. R.; assignee: Mine Safety Appliances Co. (26.01.1971). Determination of reactive hydrocarbons in air. *Patent US 3558283 A*. Filed 09.06.1967. Available: <https://www.google.si/patents/US3558283>
8. Horiba. *Explore the future.* Available: <http://www.horiba.com/ru>
9. *Gazoanalizatory i gazoanaliticheskoe oborudovanie ETEK.* Available: <http://etek-ltd.ru/>
10. Sexton, F. W., Michie Jr., R. M., McElroy, F., Thompson, V. (1982). *A Comparative Evaluation of Seven Automated Ambient Nonmethane Organic Compound Analyzers.* EPA/600/4-82/046 (NTIS PB82230798). Washington, D.C.: U.S. Environmental Protection Agency, 3.
11. Vartanov, A. Z., Ruban, A. D. (2009). *Metody i pribory kontroliia okruzhaiushchei sredy i ekologicheskogo monitoringa.* Moscow: Gornaia kniga, 640.
12. Ivasenko, V. M. (2015). Doslidzhennia osnovnykh metrolohichnykh kharakterystyk polumiano-ionizatsiinoho detektora. *Visnyk Natsionalnoho tekhnichnoho universytetu Ukrainy «Kyivskiy politekhnichnyi instytut»*. Ser. *Pryladobuduvannia*, 49, 75–81.
13. Schenck, H. (1979). *Theories of Engineering Experimentation.* Ed. 3. CRC Press, 302.
14. Lucero, D. P. (1972). Water Vapor Sensitivity Response of Hydrogen Flame Ionization Detectors. *Journal of Chromatographic Science*, 10 (7), 463–467. doi:10.1093/chromsci/10.7.463

INCREASE OF ECOLOGICAL SAFETY OF MINE WATER DISPOSAL IN THE LVIV-VOLYN COAL BASIN

page 57–63

The object of research is the process scheme with adjustable mine water disposal mode of «Lubelska» № 1–2 mine via the channel «Butynsky» in the river Rata (Ukraine), and – the impact on the most vulnerable components of the environment – surface water and groundwater of Lviv-Volyn coal basin. In connection with the change in the projected method of shaft sinking it is necessary to review and adjust the system of mine water disposal during the building period. For this work it is analyzed and summarized hydrogeochemical data obtained in the course of monitoring observations in the course of 3 years in the control supervisory sections of «Butynsky» channel and the river Rata.

As a result, the technical possibilities and environmental friendliness for adjustment of mine water disposal scheme in the second stage of shaft sinking (during building) are shown. The method of flow diagram adjustment of mine water disposal is chosen. It can be done in two modes, on the basis of the prediction of quantity and quality composition of the mine water.

The first mode – discharge in accordance with the predicted value of the mine water inflow: minimum periods of water inflow – 30–40 m³/h with a salinity of 20 g/dm³ and periods of high water inflow – 80–90 m³/h with a salinity of 10 g/dm³.

The second mode – the accumulation of mine water in the settling ponds, mixing them with drainage water formed as a result of drainage of the industrial site of «Lubelska» № 1–2 mine and with rain water, then – averaging and discharge in a volume up to 90–100 m³/h with salinity up to 5–7 g/dm³.

This disposal scheme will improve the environmental safety of mine water disposal into surface water bodies, regulate the conditions of mine water discharge, as well as their quality within the allowable concentrations and, thus, minimize the impact on the hydrochemical regime of surface water of Butynsky channel and the river Rata.

Keywords: mine water, building of coal mines, system of controlled discharge of mine water.

References

1. *Otsinka ta obgruntuvannya ekolohichno pryiniatnykh rishen z okhorony vodnoho seredovyschcha v umovakh prokhozhenia stoliv shakhty «Liubelska» № 1-2 ta rozroblennia proektu HDS v «Butynskiyi» kanal na period budivnytstva.* (2011). Kharkiv: UkrNDIEP, 116.
2. Kroll, A., Amezaga, J. M., Younger, P. L., Wolkersdorfer, C. (2002). Regulation of Mine Waters in the European Union: The Contribution of Scientific Research to Policy Development. *Mine Water and the Environment*, 21 (4), 193–200. doi:10.1007/s102300200043
3. *International Assessment of Marine and Riverine Disposal of Mine Tailings.* (2013, May). Study commissioned by the Office for the London Convention and Protocol and Ocean Affairs, IMO, in collaboration with the United Nations Environment Programme (UNEP) Global Programme of Action. Available: <http://www.imo.org/en/OurWork/Environment/LCLP/newandemergingissues/Documents/Mine%20Tailings%20Marine%20and%20Riverine%20Disposal%20Final%20for%20Web.pdf>
4. Zgorska, A., Trzaski, L., Wiesner, M. (2016). Environmental risk caused by high salinity mine water discharges from active and closed mines located in the Upper Silesian Coal Basin (Poland). *Proceedings IMWA 2016, Freiberg/Germany. Mining Meets Water – Conflicts and Solutions*, 85–92. Available: https://www.imwa.info/docs/imwa_2016/IMWA2016_Zgorska_95.pdf
5. Jarvis, A. P., Younger, P. L. (2000). Broadening the scope of mine water environmental impact assessment. *Environmental Impact Assessment Review*, 20 (1), 85–96. doi:10.1016/s0195-9255(99)00032-3
6. Matlak, E. S., Maleev, V. B. (1991). *Snizhenie zagriaznenosti shahtnykh vod v podzemnykh usloviiah.* Kyiv: Tehnika, 136.
7. In: Hahn, H. H., Klute, R. (1988). Pretreatment in Chemical Water and Wastewater Treatment. *Proceedings of the 3rd Gothenburg Symposium 1988, 1.-3. Juni 1988, Gothenburg.* Springer Berlin Heidelberg, 260. doi:10.1007/978-3-642-73819-7
8. Innovatsionnye puti modernizatsii bazovykh otraslei promyshlennosti, energo- i resursoberezhnie, ohrana okruzhaiushchei prirodnoi sredy. (2016). *Sbornik nauchnykh trudov V Mezhdunarodnoi nauchno-prakticheskoi konferentsii molodykh uchenykh i spetsialistov, 23-24 marta 2016, Kharkiv.* Kharkiv: GP «UkrN-TTs «Energostal'», 209.
9. Harionovskii, A. A. (2000). *Kompleksnaia ochistka shahtnykh i kar'ernykh vod ot tehmogennykh zagriaznenii.* Shahty: YuRO AGN, 3–26.
10. Pro poriadok rozroblennia i zatverdzhennia normatyviv hranychno dopustymoho skydannia zabrudniuiuchykh rehovyn ta perelik zabrudniuiuchykh rehovyn, skydannia yakykh normuietsia. *Decree of the Cabinet of Ministers of Ukraine from 11.09.1996 № 1100.* Available: <http://zakon3.rada.gov.ua/laws/show/1100-96-п>
11. Mongait, I. L., Tekinidi, K. D., Nikoladze, G. I. (1978). *Ochistka shahtnykh vod.* Moscow: Nedra, 173.
12. Cherkinskii, S. N. (1971). *Sanitarnye usloviia spuska stochnykh vod v vodoemy.* Moscow: Stroizdat, 208.
13. Merkulov, V. A. (1981). *Ohrana prirody na ugol'nykh shahtah.* Moscow: Nedra, 184.